

Materials for Generation IV Nuclear Reactors

NATO Advanced Study Institute, Cargese, Corsica, France September 24 - October 6, 2007

Research and Development of Oxide Dispersion Strengthened Ferritic Steels for Sodium Cooled Fast Breeder Reactor Fuels





October 4, 2007

Masaki INOUE*, Takeji KAITO, Satoshi OHTSUKA

Core and Structural Materials Group,
FBR System Technology Development Unit,
Advanced Nuclear System Research and Development Directorate,
Japan Atomic Energy Agency (JAEA)



Contributors and Collaborators

Hokkaido University
Professor Shigeharu UKAI

Kobelco Research Institute, Inc.

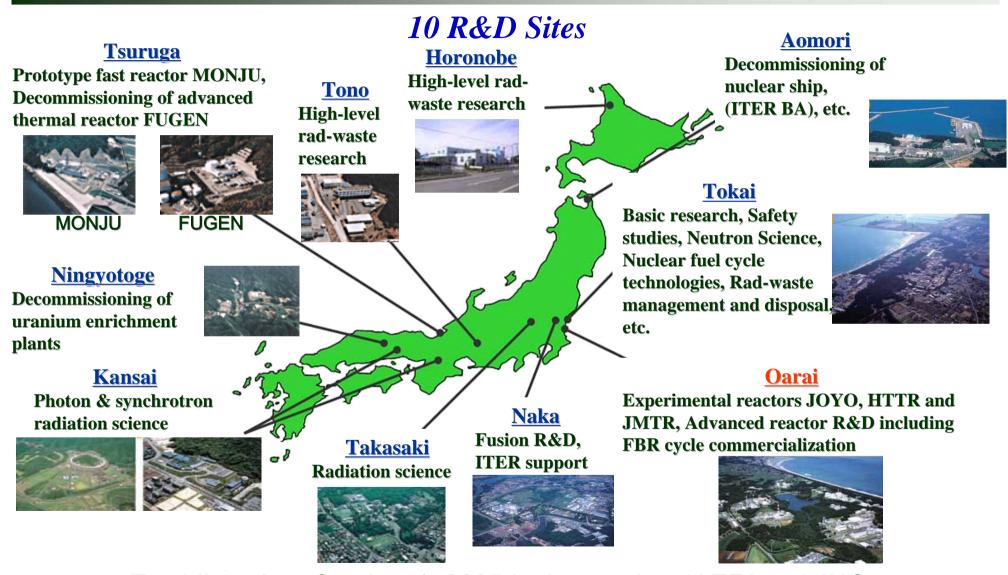
Dr. Masayuki FUJIWARA

Dr. Takanari OKUDA

Sumitomo Metal Technology, Inc. Mr. Toshimi KOBAYASHI



Japan Atomic Energy Agency (JAEA)



Established on October 1, 2005 by integrating JAERI and JNC Annual Budget: ~200BJPY, Employees: ~4,400



Outline

- 1. Fast Reactor Cycle Development in Japan
- 2. Introduction to ODS Steels
- 3. Alloy Design

Oxide Dispersion Strengthened: ODS

- 4. Microstructure Control (Nano-Size Oxide Particles) (Grain Morphology)
- 5. Mechanical Properties
- 6. Irradiation Tests
- 7. Future Mass Production
- 8. Conclusions

Chapter 1. FR Cycle Development in Japan

- "Feasibility Study" on Commercialized Fast Breeder Reactor Cycle Systems was conducted from JFY1999 to 2005.
- The final reports were compiled in March 2006, and reviewed by the Government (AEC, MEXT, METI).
- The Most Promising Concepts to Commercialize FR Cycle

Reactor: Sodium-cooled Fast Reactor with MOX fuel

Fuel Cycle: Advanced Aqueous Reprocessing

Simplified Pelletizing Fuel Fabrication

•FR cycle technology development has advanced from "Feasibility Study" to "Project" since 2006.

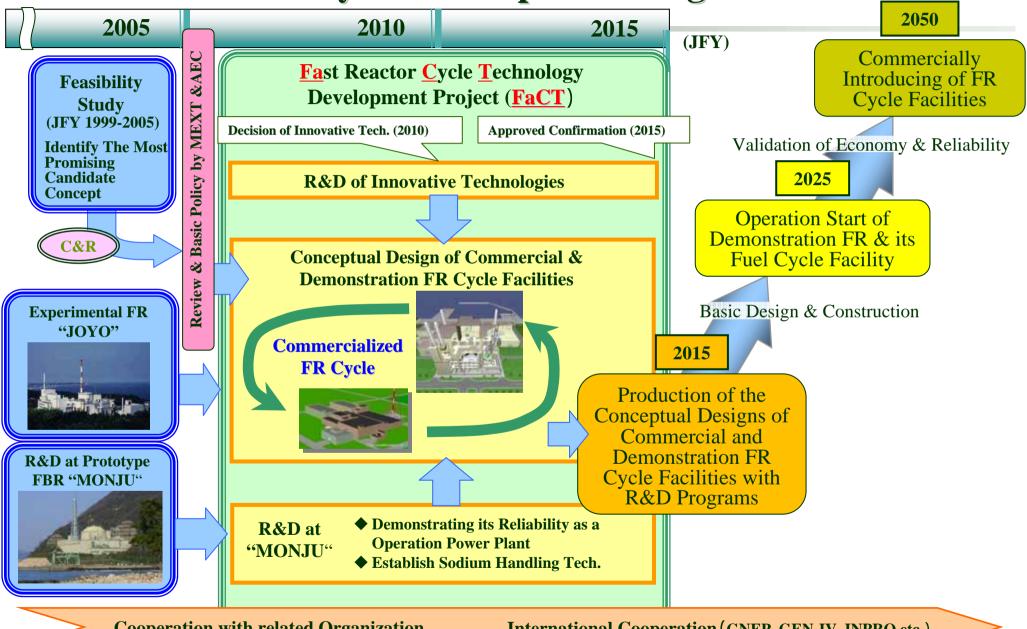
AEC: The Japan Atomic Energy Commission

MEXT: The Ministry of Education, Culture, Sports, Science and Technology

METI: The Ministry of Economy, Trade and Industry



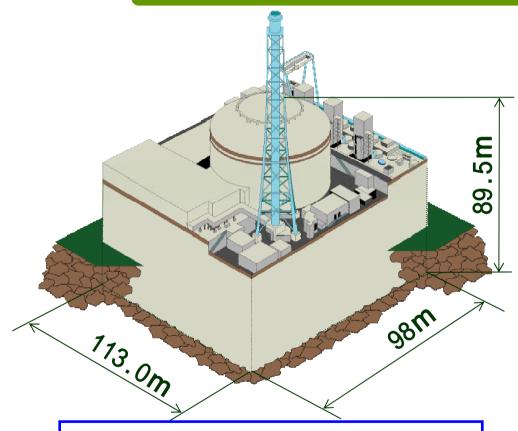
FR Cycle Development Program





Comparison of Reactor Size

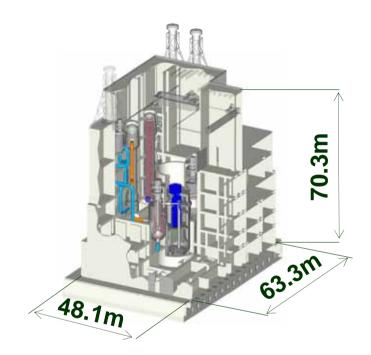
Electricity output of JSFR is about 5 times larger while site area is about 1/4.



Prototype FBR MONJU

Thermal Output: 714 MWt

Electricity Output: 280 MWe



Japanese Sodium cooled

Fast Reactor (JSFR)

Thermal Output: 3,570 MWt

Electricity Output: 1,500 MWe



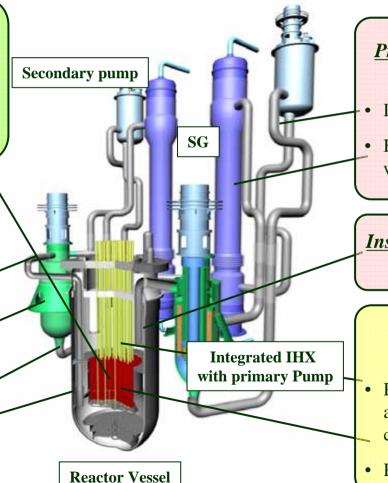
<u>Japanese Sodium-cooled Fast Reactor (JSFR)</u>

- ▶ 1,500 MWe large-scale SFR with MOX fuel,
- Innovative technologies for enhancement of reactor core safety, high economic competitiveness and countermeasures against specific issues of sodium

ODS Steels for cladding tube to achieve high burn-up at elevated temperatures

Innovative technologies to reduce
plant materials and reactor building
volume

- Two-loop cooling system
- Shortening of piping with high chromium steel
- Integrated Pump-IHX Component
- Compact reactor vessel



Prevention of sodium chemical reactions

- Double-wall piping
- High reliable SG with doublewall tube

Inspection and repair technology under sodium

Enhancement of reactor core safety

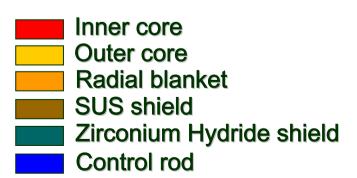
- Passive reactor shutdown system and decay heat removal by natural circulation
- Re-criticality free core



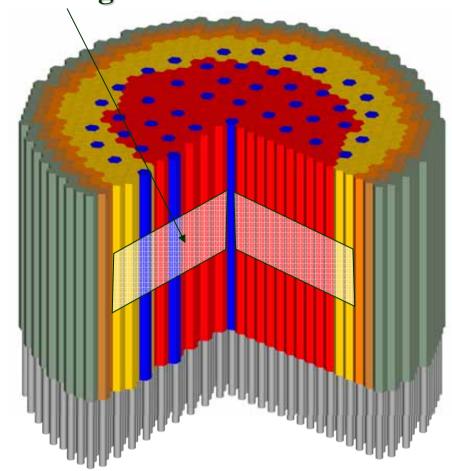
Reactor Core Configuration of JSFR

Neutronic Characteristics

Output power (MWt)	3570		
Cycle length (months)	26		
Refueling batch [core/RB]	4/4		
Pu-enrichment (wt%) [Inner/outer]	18.3/20.9		
Burnup reactivity (%dk/kk')	2.3		
Breeeding ratio	1.10		
Discharge burnup (GWd/t) [core]	147		
[core + blanket]	90		
Pu fissile inventory (t/GWe)	5.7		
Maximum neutron dose (dpa)	250		
Sodium void reactivity (\$)	5.3		



Core region

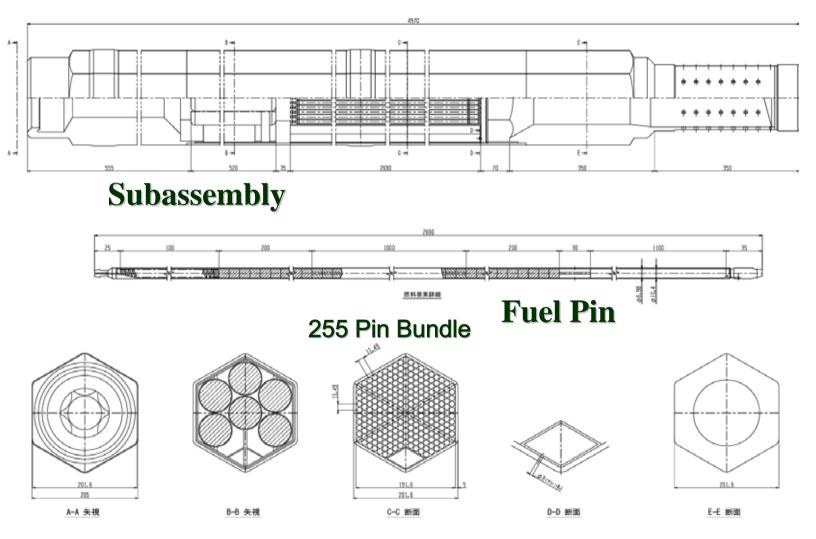


Cut view of reactor core

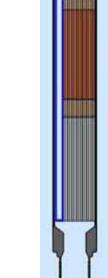


A Drawing of Core Fuel Pin and Subassembly

An example of design study for 1.5GWe SFR cores



Molten fuel discharge duct at a corner of wrapper tube



The original figure can be found in JAEA Research 2006-042 (2006), P.249



Chapter 2. Introduction to ODS Steels

Crystallographic Consideration against Irradiation Damages

- Body Centered Cubic > Face Centered Cubic

Requirements and Target Performance for ODS Steels

- Reactor and Fuel Cycle
- Fuel Pin Mechanical Design

Manufacturing Process

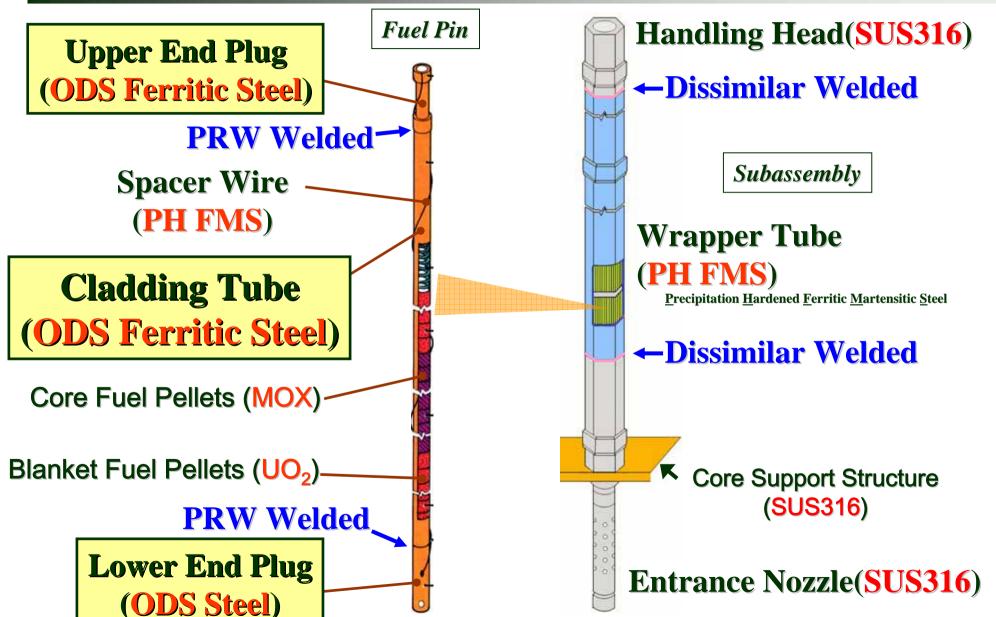
- Powder Metallurgy
- Precision Seamless Tube Production

Break Through (Historical Review)

- Microstructure Control
- Pilger Cold Rolling Process

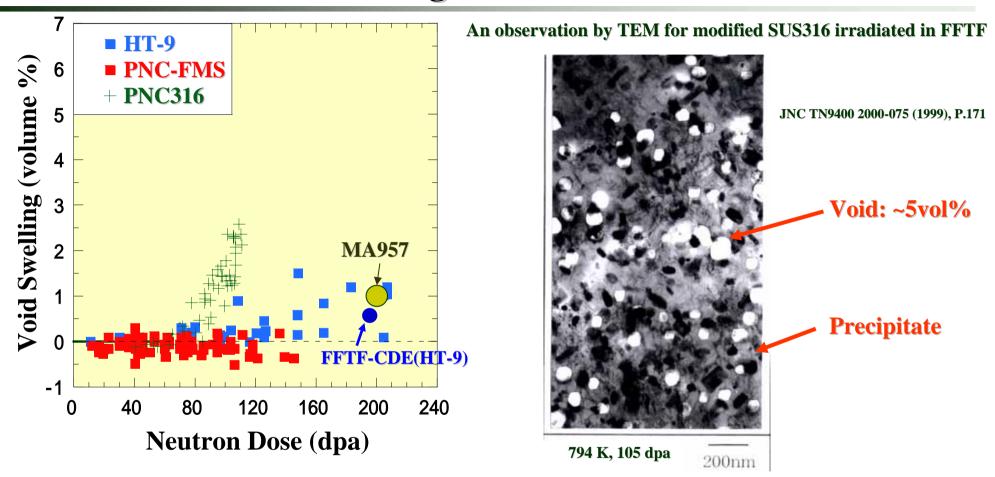


Materials for Core Fuel Pin and Subassembly





Void Swelling in Various Steels

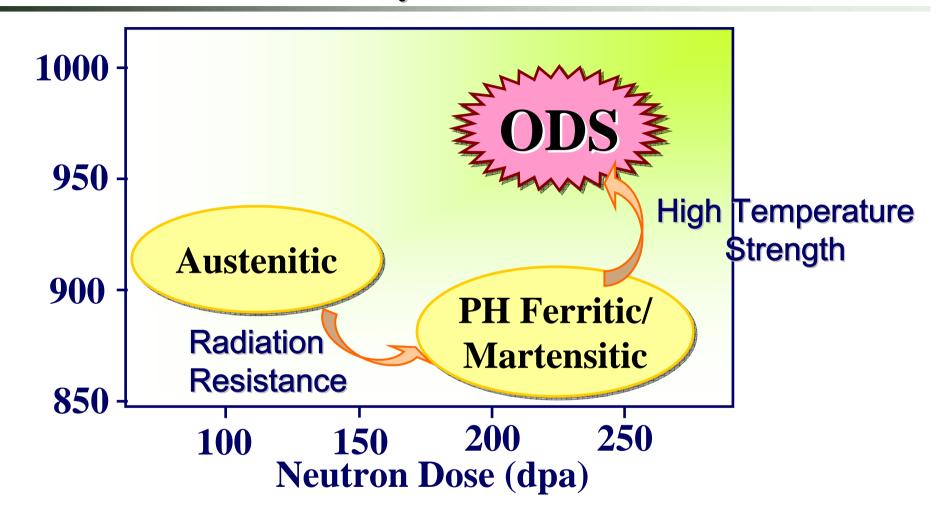


Void swelling will be detrimental in austenitic stainless steels over 120 dpa. BCC exhibits better swelling resistance than FCC.

Ferritic steels are promising for higher doses than 120 dpa.



Why ODS?



Precipitation hardening will be lost in ferritic steels over 923 K.

Oxide dispersion strengthening will be effective even over 973 K.

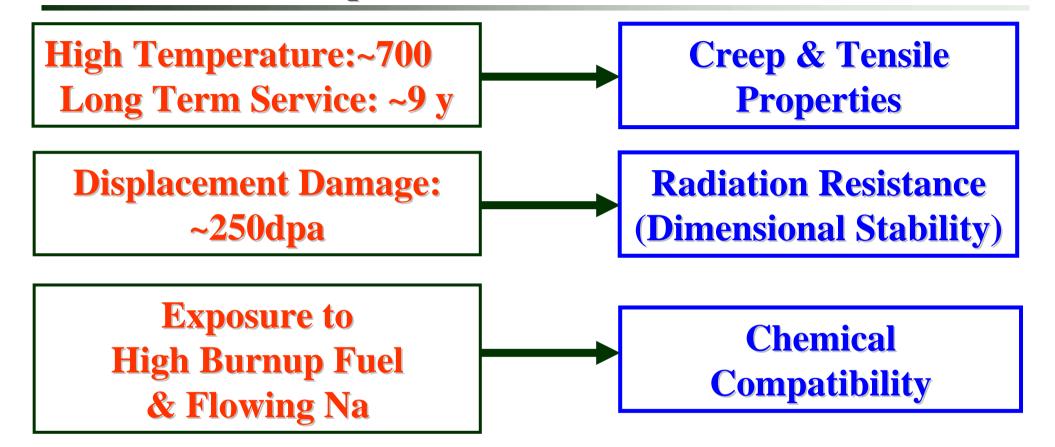


Requirements for Fuel Pin Cladding Tubes

- 1. Power Plant Performance: Higher Thermal Efficiency Reactor Outlet Coolant Temperature: 823 K (550) Reactor Inlet Coolant Temperature: 673 K (400)
- Maximum (Hot Spot) Temperature: 973 K (700°C)
- 2. Fuel Cycle Performance: Higher Burnup
 Discharge Average Burnup of Core Fuel: 150 GWd/t
 Long Term Service: ~9 years (26 months × 4 Batches)
- Peak Neutron Dose: ~250 dpa
- Peak Burnup: ~250 GWd/t
- Maximum Internal Pressure: ~12 MPa



Requirements for ODS Steels



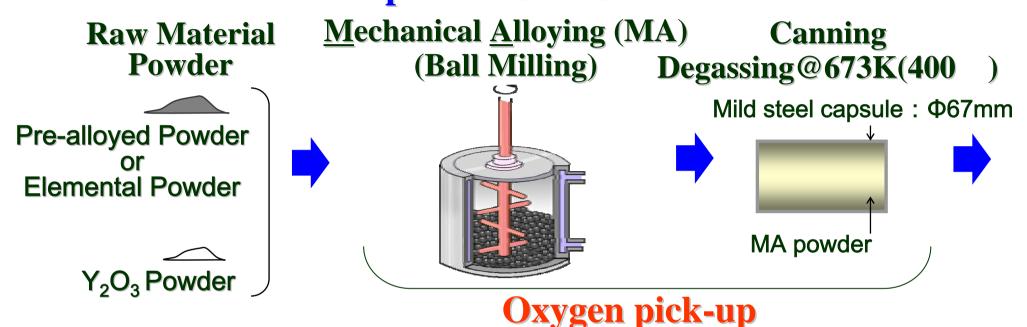
Mechanical Properties are targeted to be

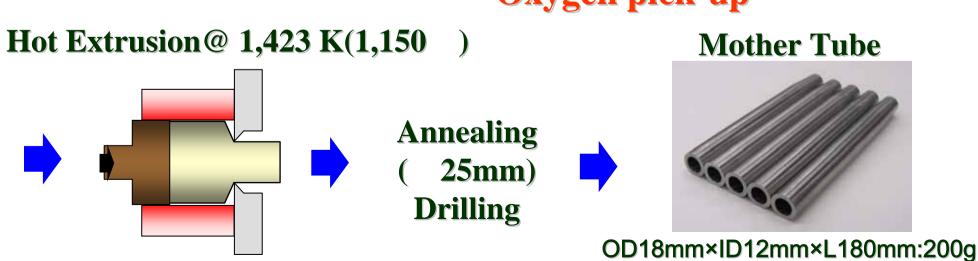
Ultimate Tensile Strength (UTS): >300 MPa @ 973 K
Internal Creep Rupture Strength: >120 MPa × 10⁴@ 973 K hr
Uniform Elongation (UE): >1%

Manufacturing Process:1 Powder Metallurgy (PM) Process for Mother Tubes



Dispersoid Size Control







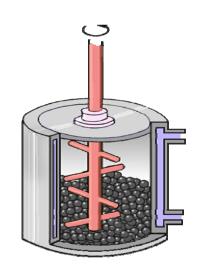
High Energy Attrition Type Ball Mill for Mechanical Alloying

10D Attritor





Balls: 150 kg Powders: 10kg/Batch 99.9999%Ar 220rpm × 48hrs



Mechanically alloyed powders are exposed to air in recovery.

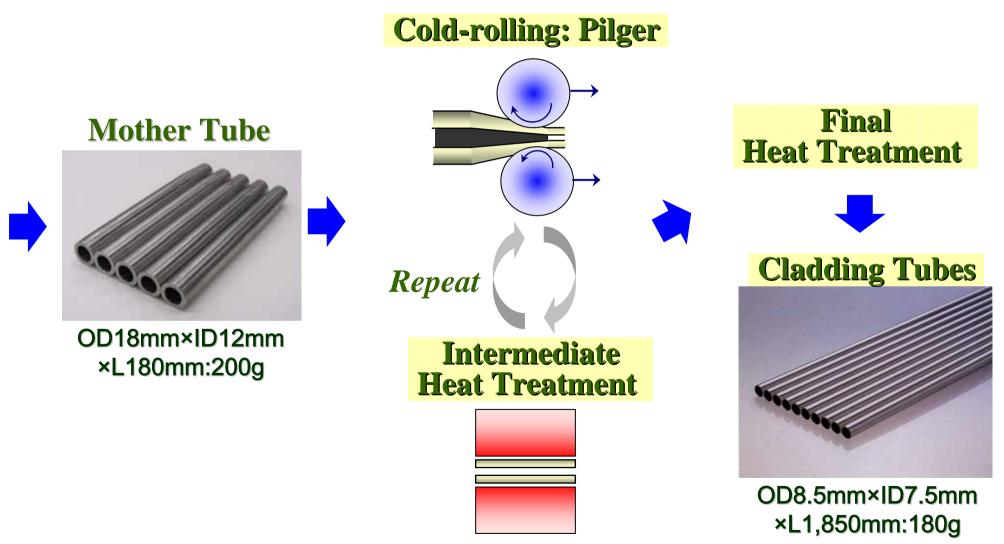
Photographs are supplied by Dr. Fujiwara of Kobelco Research Institute, Inc.

Manufacturing Process:2



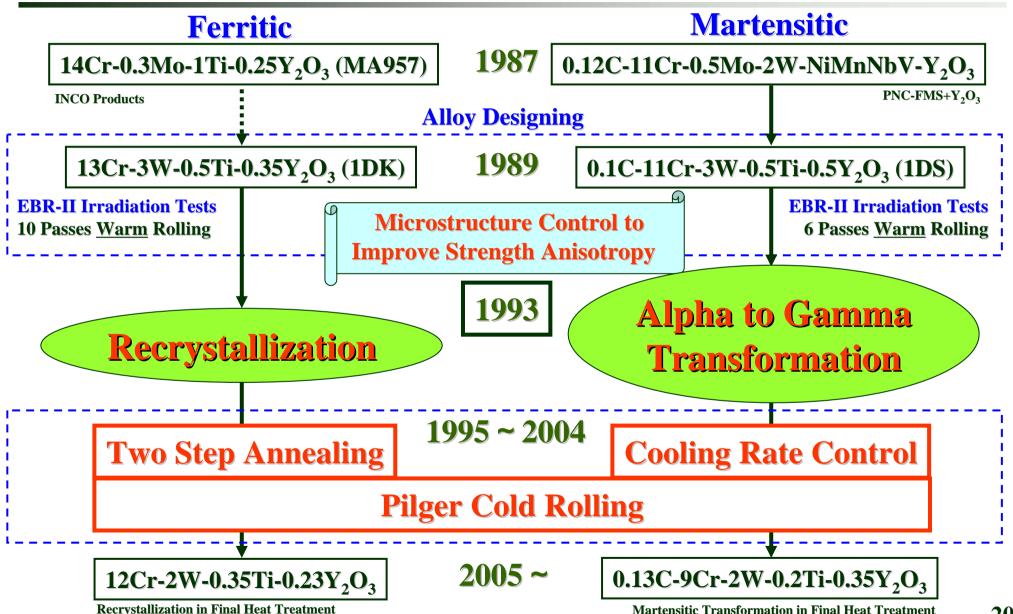
Cold Rolling Process from Mother Tubes to Cladding Tubes

Grain Morphology Control





History and Major Break Through



4 Passes Cold Rolling

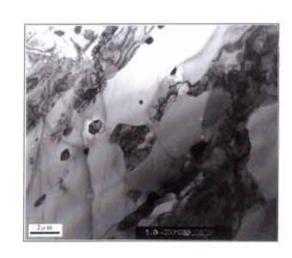


Two Candidates of ODS Steels for Further R&D in FaCT

- 1. Primary Candidate: 9Cr-ODS
- Fe-0.13C-9Cr-2W-0.2Ti-0.35Y₂O₃
- Normalizing@1,323 K(1,050) × 60min Tempering@1,053-1,073 K × 60min
- Tempered Martensitic Matrix
- Alpha to Gamma Transformation
- Cooling Rate Control



- 2. Secondary Candidate: 12Cr-ODS
- Fe-0.03C-12Cr-2W-0.26Ti-0.23Y₂O₃
- -Annealed@ \sim 1,423K(1,150) × \sim 60min
- Recrystallization
- Two-step Annealing





Chapter 3. Alloy Design

Selecting alloying elements

- Understanding strengthening mechanisms
- Predicting phase diagrams

Optimizing chemical compositions

- Specimen preparations
- Tensile and creep tests
- Trade-off between strength, toughness and manufacturing

Unanticipated phenomena and empirical approach

- phase formation (not equilibrium phase) in 9Cr-ODS
- Recrystallization (highly empirical) in 12Cr-ODS



in wt%

Alloying Concepts and Specifications

Element	C	Cr	W	Ti	Y_2O_3	Ex.O
Mechanism	M	M	S	D	D	D
9Cr-ODS	0.13	9.0	2.0	0.20	0.35	0.07
12Cr-ODS	0.03	12.0	2.0	0.26	0.23	0.07

Impurity: Mn, Ni, N, Ar, Si, P, S Ex.O = TolalOxygen Content $-\frac{48}{178}$ Yttrium Content

Oxygen picked up during PM process is called as Excess Oxygen (Ex.O), but it is the most important alloying element !!

D: Dispersion Strengthening, S: Solution Hardening,

P: Precipitation Hardening, M: Phase Control

Martensitic/Ferritic & Precipitation Hardened: PNC-FMS

Element	C	Cr	Mn	Ni	Mo	W	V	Nb	N
Mechanism	M/P	M	M	M	S	S	P	P	P
PNC-FMS	0.12	11.0	0.60	0.40	0.50	2.0	0.20	0.05	0.05



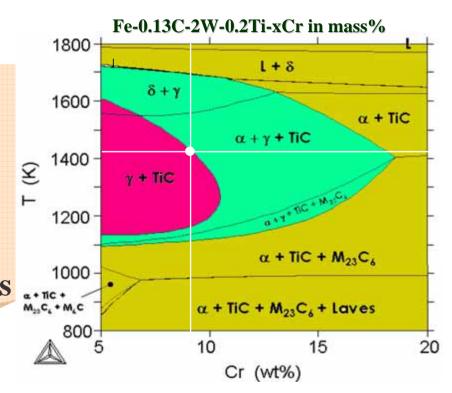
Cr Contents: Phase Control and Chemical Compatibility

• <u>9Cr</u>-ODS Steel: Martensitic

- Large capacity for trapping radiationinduced point defects.

Lower irradiation embrittlement

-Alpha to Gamma Transformation
Anisotropy-free mechanical properties

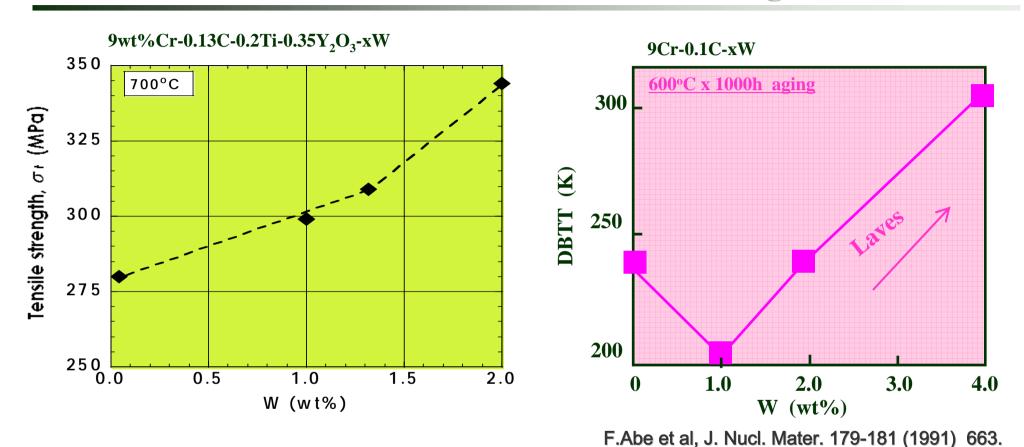


• 12Cr-ODS: Fully Ferritic

- More stable oxide films on surface
Fuel-cladding chemical interaction (FCCI)
Nitric acid dissolution
Better corrosion resistance with minimal embrittlement

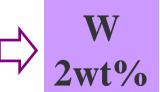


W Contents: Solution Hardening



- Increasing to 2wt% improve considerablely high temperature strength.

- Excess addition over 2wt% will provide a concern with irradiation/thermal embrittlements due to laves precipitation.





Oxide Dispersion Strengthening Theory

Smaller obstacles (dispersoids) interspacing provides

- higher threshold stress to dislocation motion and
 - higher strength at elevated temperature.

Finer dispersoids and higher number density is desirable.

Threshold stress to Dislocation Motion

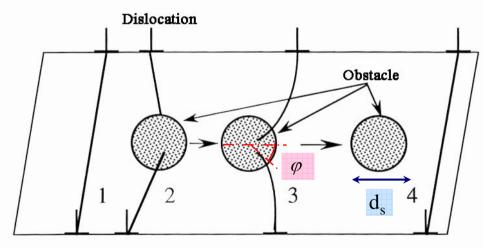
$$\tau_{Or} = A_{V} \frac{Gb}{2\pi\overline{\lambda}} \left[\ln \left(\frac{\widetilde{D}}{r_{0}} \right) + B_{0} \right]$$

•
$$\frac{1}{\widetilde{D}} = \frac{1}{\overline{\lambda}} + \frac{1}{\overline{d}_s}$$
 $\overline{\lambda}$: Average obstacle spacing \overline{d}_s : Average obstacle diameter

 \bullet r₀: dislocation core cut-off radius

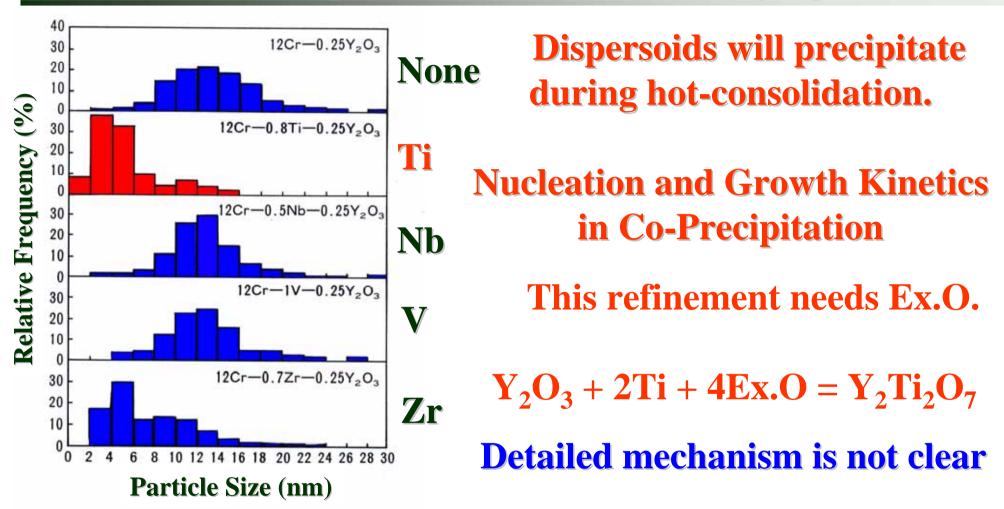
 $\bullet A_{V} = \begin{cases} \frac{1 + v \sin^{2} \varphi}{1 - v} \cos \varphi & (\varphi = 47^{\circ}), B_{0} = 0.6 \\ \left(1 - \frac{v}{1 - v} \sin^{2} \varphi\right) \cos \varphi & (\varphi = 19^{\circ}), B_{0} = 0.7 \end{cases}$: Edge dislocation

✓ Srolovitz mechanism: Incoherent obstacle





Refinement of Dispersoid Size by Minor Alloying Elements



Dispersoid sizes depend on minor alloying element(s) and amount of Ex.O.

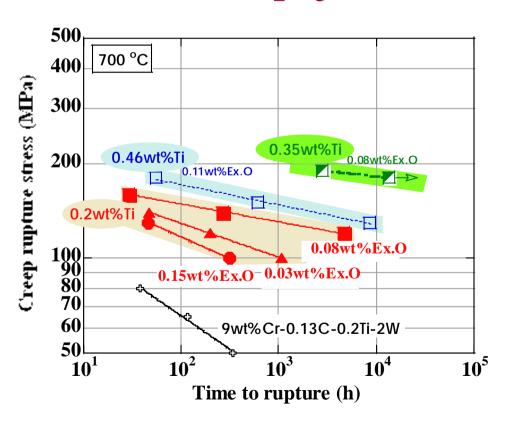
Titanium addition with Ex.O is very effective to enhance ODS effect. 27

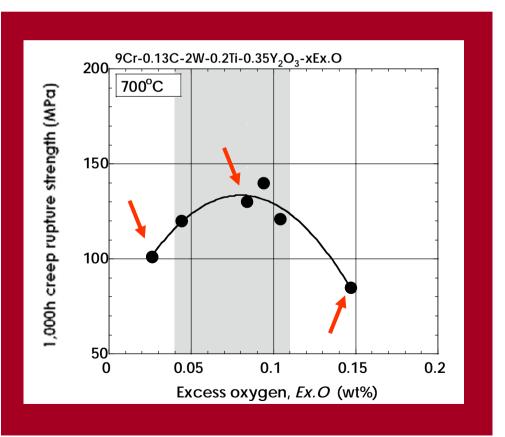


Y₂O₃, Ti & Excess Oxygen (Ex.O) Contents

Increasing Y_2O_3 and Ti contents improve mechanical strength, while it leads to considerable degradation of tube manufacturability. Too much Ex.O degrades mechanical strength.

 $=> 0.2 \text{Ti-} 0.35 \text{Y}_2 \text{O}_3 - 0.08 \text{Ex.O}$

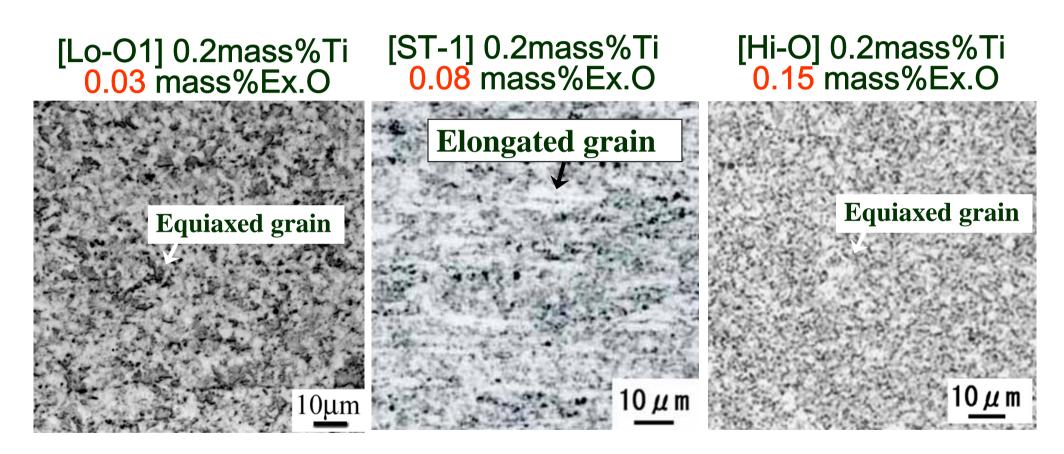






Phase Characterizations for the 9Cr-ODS Steel

phase disperses in martensitic with certain amount of Ex.O.

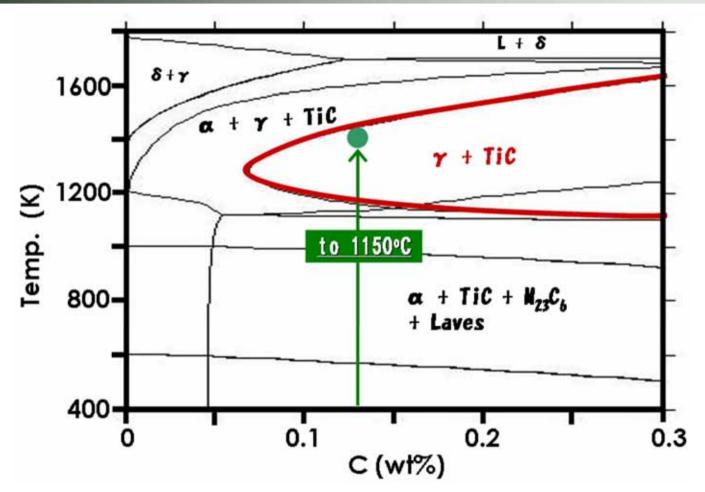


Equiaxed Grain: Martensite

Elongated Grain: ferrite



Unanticipated Phenomena in the 9Cr-ODS Steel

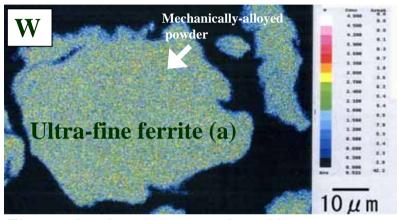


Phase diagram prediction does not indicate ferrite formation.

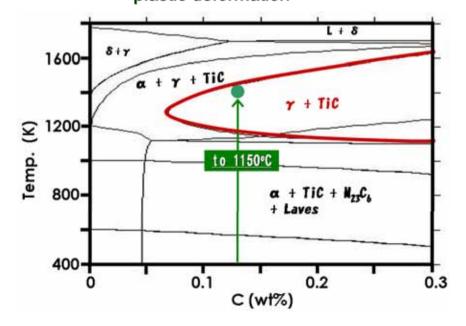


Ferrite Formation during Hot Consolidation

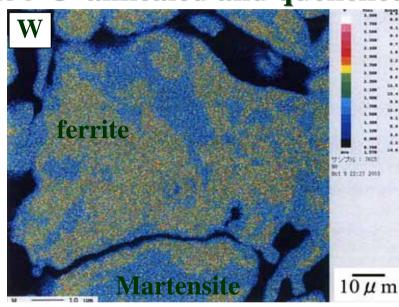
As mechanically alloyed



Fine-grained structure produced by severe plastic deformation



●1150°C annealed and quenched



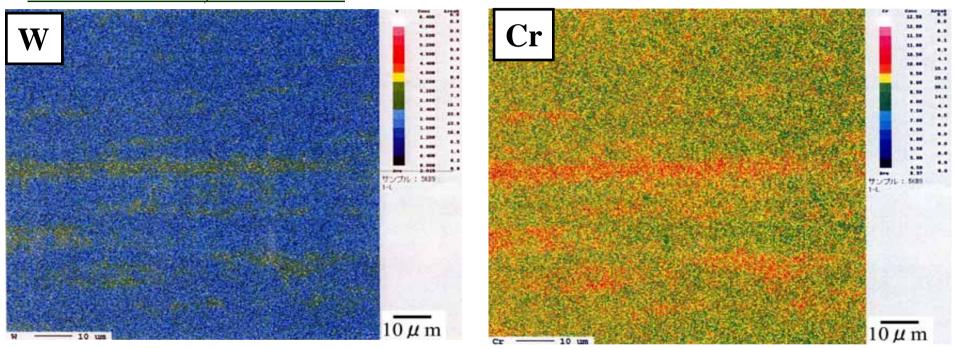
Ferrite()/austenite() duplex

ferrite Formation
Non-equibrium phase at 1150°C.



EPMA Mapping of Longitudinal Section for the 9Cr-ODS Steel

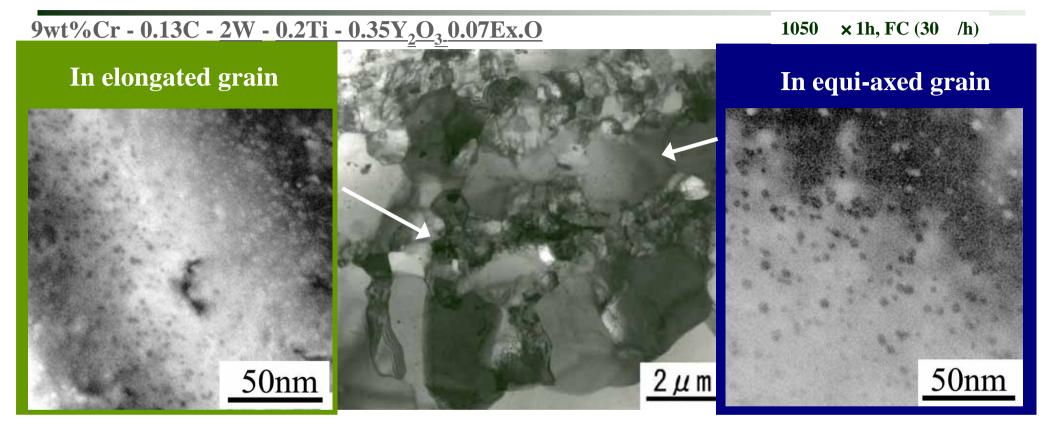
●Ex.O=0.07wt%, Ti=0.2wt%



- ●Elongated grains (), where W and Cr are concentrated, exist in the matrix of high strength steel.
- •Elongated grains can be also called as residual -ferrite, which remained untransformed during 1150°C hot-consolidation.



TEM Observation for the 9Cr-ODS Steel

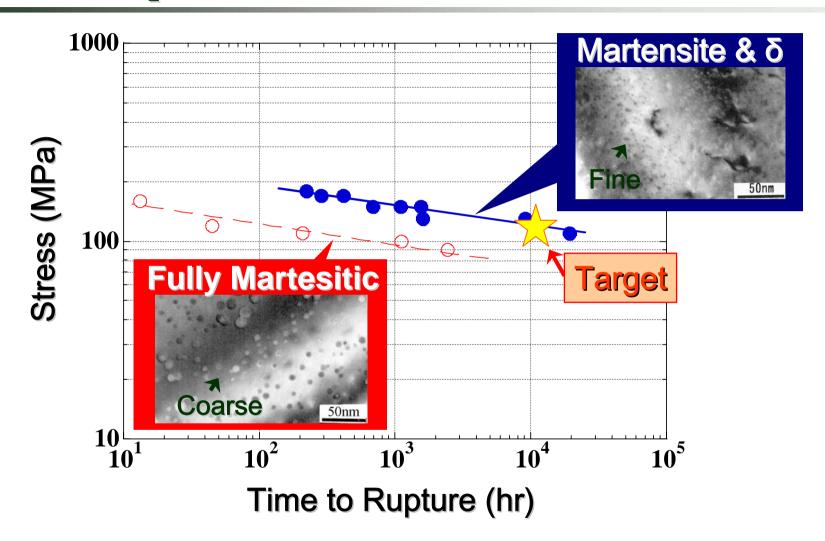


- Oxide particles is more finely dispersed in elongated grain () than in equiaxed grain.
- The reason why the particle distribution between and martensite* are different is not clear.

^{*:} Martensite should be gamma phase during hot-extrusion and heat treatments.



Composite Nature of the 9Cr-ODS Steel



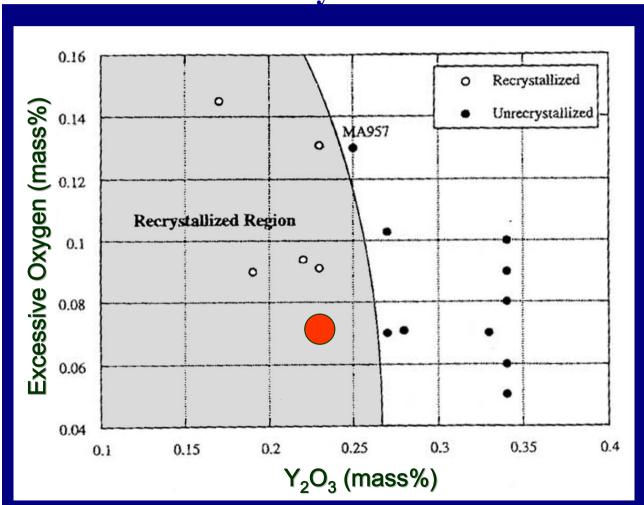
- Certain amount of ferrite is essential to attain the target.
- The 9Cr-ODS steel is re-enforced by ferrite.



Recrystallization Threshold in the 12Cr-ODS Steel

Too much Ex.O addition produces excessive, since Y-Ti-O particles are detrimental to recrystallization control.

=> 0.3Ti-0.23Y $_2$ O $_3$ => 0.07Ex.O





Summary of Alloy Effects

	Improvement	Degradation
	✓ Corrosion Resistance	✓Irradiation/Thermal Embrittlement
Cr	(<u>≥</u> 12 mass%)	(Cr-rich phase: α')
		√δ-ferrite in martensite
	✓ High-temperature Strength	✓Irradiation/Thermal Embrittlement
W	(Solution Hardening)	(Laves, d-ferrite in Martensite)
	(Finer Precipitation of M ₂₃ C ₆)	
Ti	✓ High-temperature Strength	✓Tube Manufacturability
Y ₂ O ₃	(ODS)(Dispersoid Morphology)	(Microstructure Controllability)
Ex.O	✓ Non-equilibrium Phase Generation	

Excess Oxygen (Ex.O) is the most important alloying element, but its effect is not well clarified.



Concepts of Alloy Design

•9Cr-ODS steel: 9Cr-0.13C-2W-0.2Ti-0.35Y₂O₃-0.07Ex.O

- C,Cr: Radiation-resistant martensitic matrix

(microstructure control)

- W: Solution hardening, little laves precipitation

- Ti, Y₂O₃, Ex.O: Oxide dispersion strengthening ferrite formation

●12Cr-ODS steel: 12Cr-0.13C-2W-0.26Ti-0.23Y₂O₃-0.07Ex.O

- Cr: Corrosion resistant fully ferritic

- W: Solution hardening, little laves precipitation

- Ti, Y₂O₃, Ex.O: Oxide dispersion strengthening Recrystallization



Chapter 4. Microstructure Control

Powder Metallurgy Process

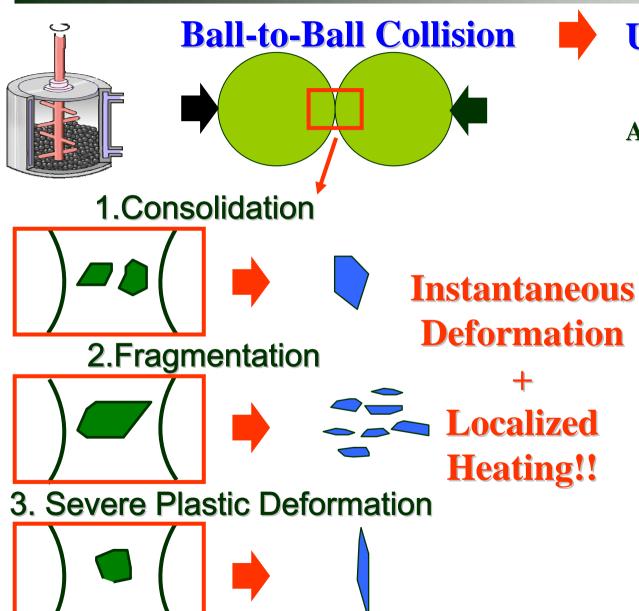
- Dispersoid Size/Distribution Control
- Uniform Solid Solution by Mechanical Alloying
- Precipitation of Nano Particles during Hot Consolidation

Precision Seamless Tube Production Process

- Grain Morphology Control
- Pilger Cold Rolling (Zircalloy Tubing)
- Intermediate and Final Heat Treatments
- Temperature History

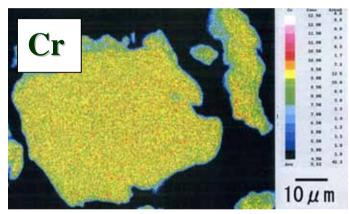


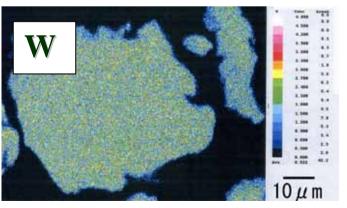
High Energy Attrition Type Ball Mill for Mechanical Alloying



Uniform Solid Solution

EPMA Mapping for As-mechanically Alloyed Powder





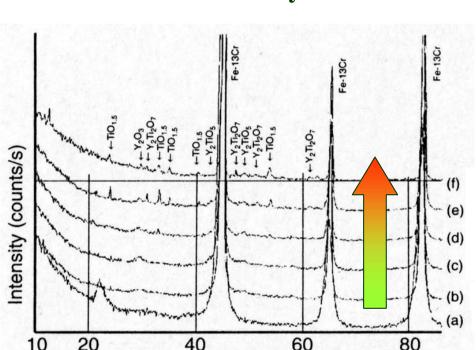
9Cr-0.2Ti-0.08Ex.O



Precipitation of Nanometer Size Oxide Particles

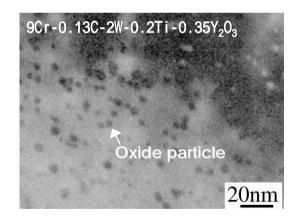
$$Y + Ti + O$$
 $Y_2O_3 \cdot yTiO_{x,}$
 $Ex.O = x \cdot y$

Yttrium oxide dissociates and dissolves into matrix during mechanical alloying due to high energy ball milling.



Titanium and yttrium complex oxide nano-particles precipitate during hot consolidation process.

(f) 1300 x 1hr (e) 1200 x 1hr (d) 1100 x 1hr (c) 1000 x 1hr (b) 900 x 1hr (a) As-MAed



The original figure can be found in "T.Okuda and M. Fujiwara, J. Materials Science Letters, 14(1995)1600".

2θ (degrees)

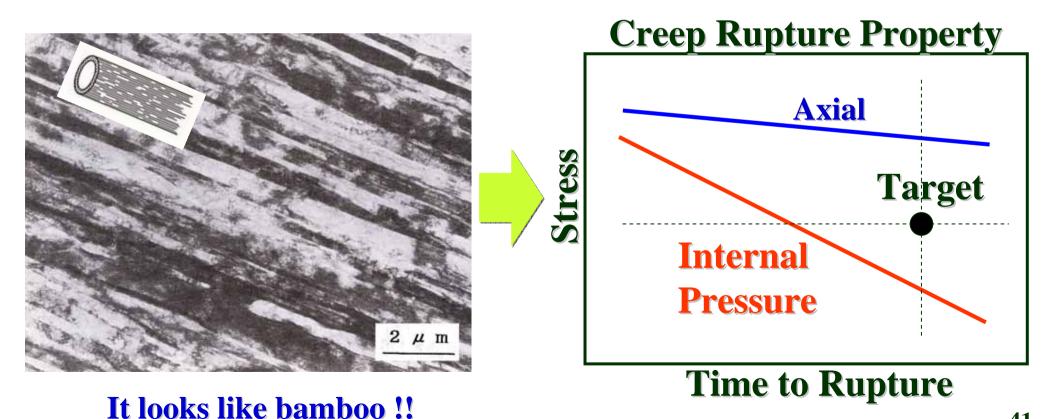
Mechanical Alloying and Hot Consolidation Dispersoid Size Control

Grain Morphology Anisotropy Results in Poor Creep Rupture Strength under Internal Pressure

(JAEA)

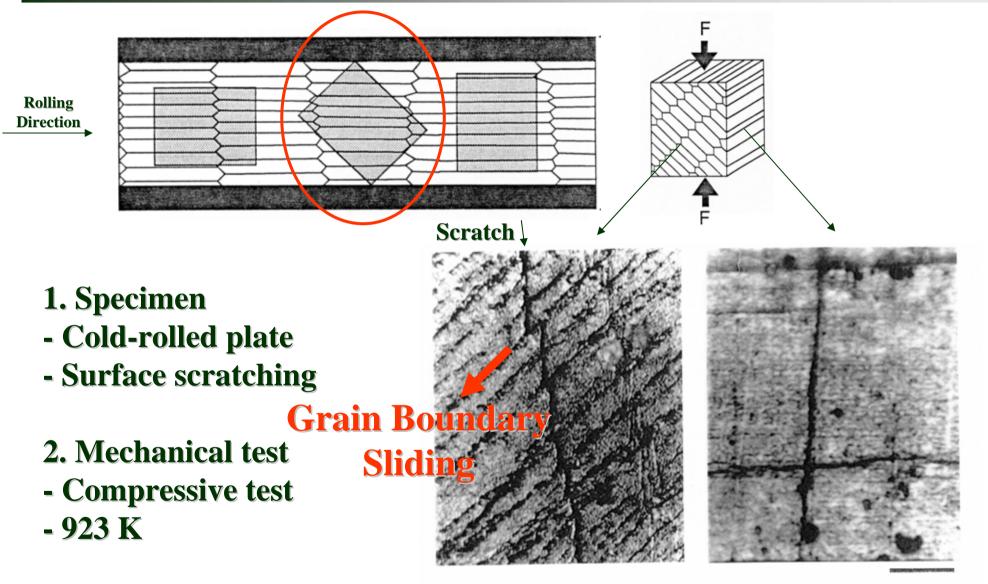
Finely elongated grain structure along rolling direction results in

- degradation of internal creep rupture strength and
- significant loss of ductility around 673 K





Grain Boundary Easily Slides at Elevated Temperature

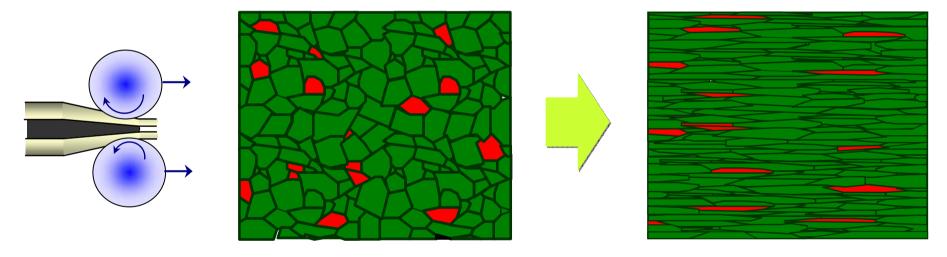




What happens in cold rolling?

Grains are elongated by uni-axial deformation Strength anisotropy

Needs: Grain morphology modification !!



Strain hardening due to cold work (area reduction: ~50%)

Hardness increase (unacceptable further cold-rolling>Hv400)

Needs: Intermediate heat treatment for softening!!

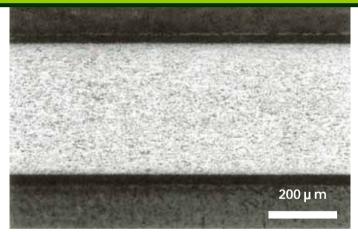


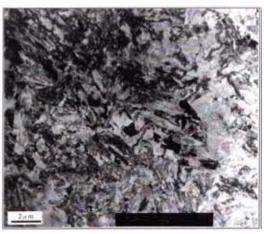
Grain Morphology in Final Products

9Cr-ODS Steel

Fe-0.13C-9Cr-2W-0.20Ti-0.35Y₂O₃

/ phase transformation

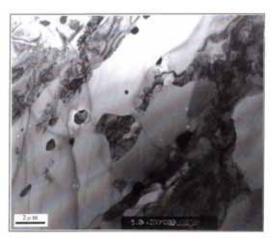




12Cr-ODS Steel

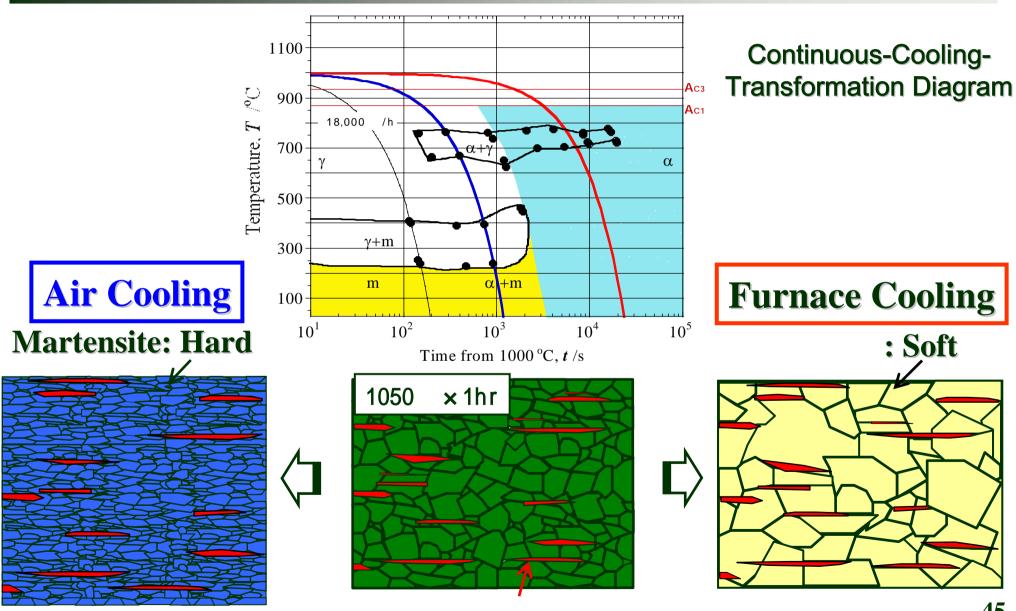
Fe-0.03C-12Cr-2W-0.26Ti-0.23Y₂O₃



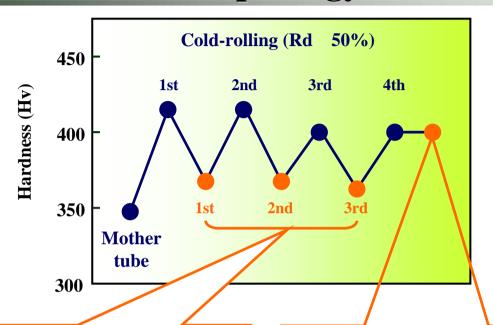




Cooling Rate and Subsequent Phases in the 9Cr-ODS Steel



Simultaneous Control of Softening/Hardening (Intention and Grain Morphology for the 9Cr-ODS Steel

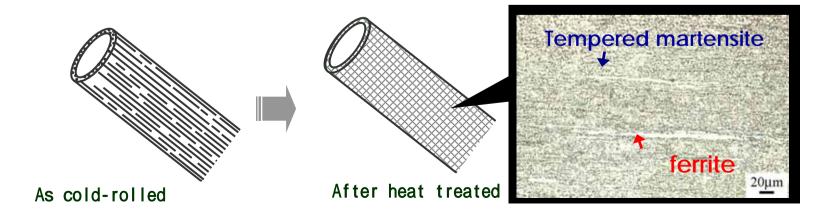


Intermediate Heat-treatment

1050°C×30 min×FC (150 °C/h)

Final Heat-treatment

1050°C×60 min×AC + 800°C×60 min×AC





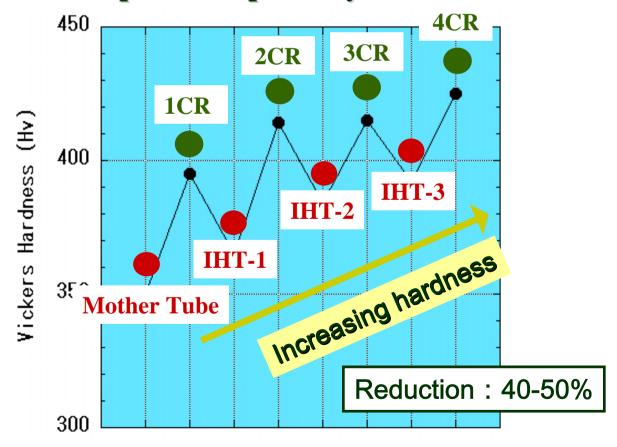
Technical Problems of the 12Cr-ODS Steel

Control of recrystalization is more difficult than that of alpha to gamma transformation.

Recrystallized grains should be produced at the final heat-treatment to improve strength and ductility against internal pressure.

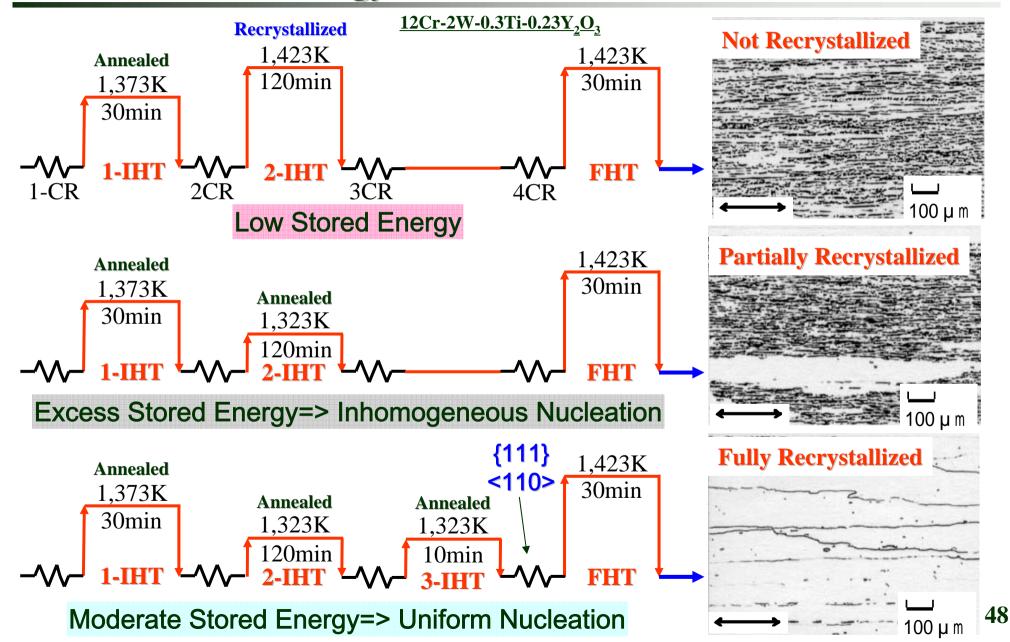
Recrystallization phenomena depends empirically on texture

evolution {111}<110>.





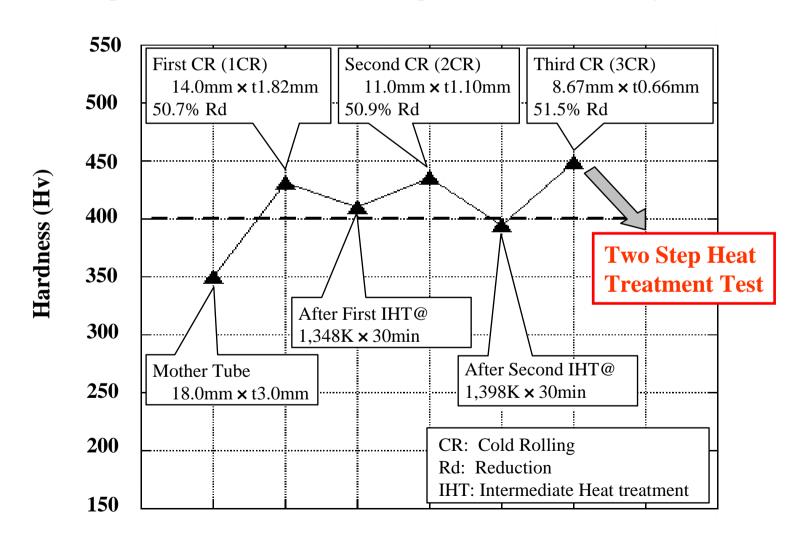
Effect of Stored Energy for Nucleation in the 12Cr-ODS Steel





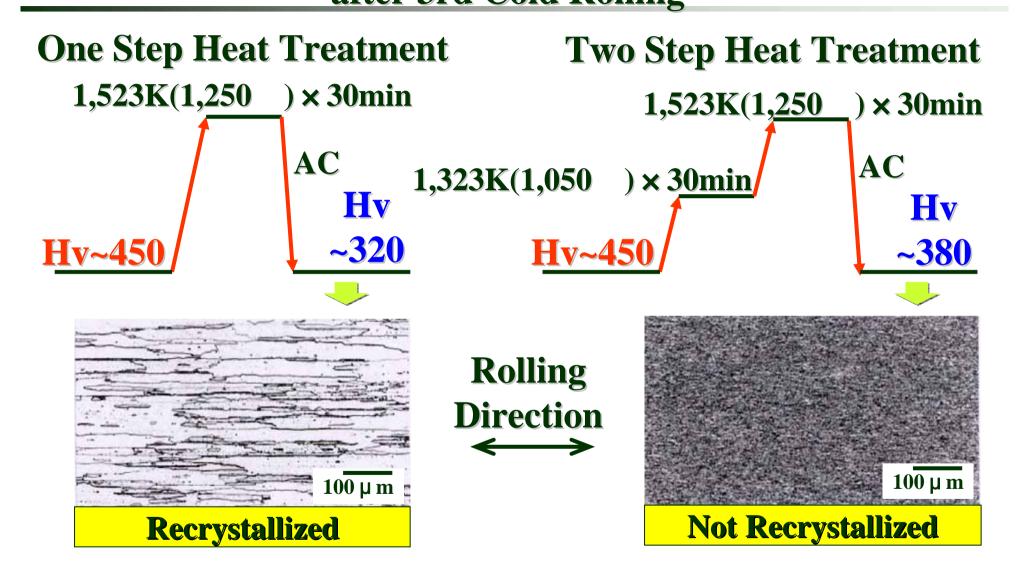
Specimen for Two-Step Heat Treatment Test

Preparation for 3rd Cold-Rolled Specimens without Recrystallizaation



One- and Two-Step Heat Treatment Tests after 3rd Cold Rolling

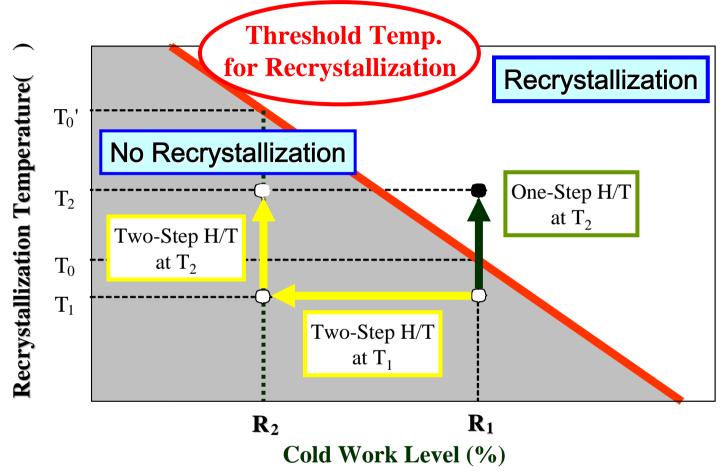




The aim of this test is to establish a heat treatment technique to consistenstly and reiliably produce the recovered structure (<400Hv) without recrystallization.

Basic idea of Two Step Heat Treatments for Recrystallization Control





The first step H/T provides the recovery of strain energy which elevates the threshold temperature for recrystallization.

The second step H/T at higher temperature can sufficiently soften cladding tube without recrystallization.



Grain Morphology Control

- •9Cr-ODS Steel: Alpha to Gamma Transformation.
 - Intermediated heat treatment with slower cooling rate softens and allows to cold roll without cracking.
 - Final heat treatment (normalizing + tempering) produces equiaxed tempered martensite (with phase) with little strength anisotropy.
- •12Cr-ODS: Recrystallization after Two-step Annealing
 - Once recrystallization took place in the course of coldrolling process, recrystallized microstructure can not be obtained at final heat treatment.
- "Two-step annealing" is useful to soften the cold-rolled tubes without premature recrystallization in intermediate heat treatments.



Chapter 5. Mechanical Properties

Deformation Mode

Steady State Operation

Noble Fission Gas Accumulation

Internal Gas Pressure Increase

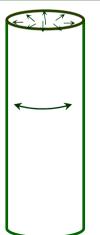
Creep Rupture Property (Long Term:~9 years) (Max. ~12 MPa)

Off-Normal Operation

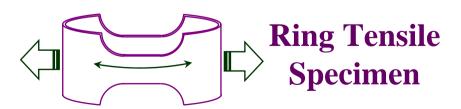
Over Power Transient

Fuel Pellet Expansion (Fuel-to-Cladding Mechanical Interaction)

Tensile Property (Short Term)



Pressurized Tube Specimen





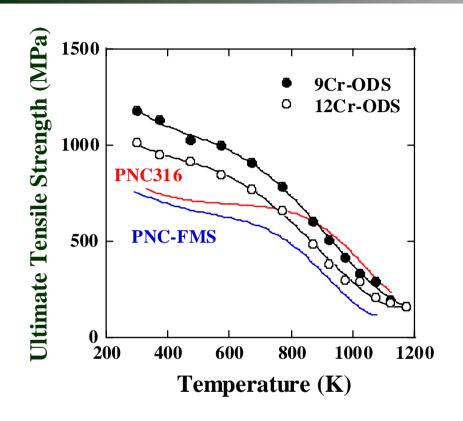
Material Strength Standard for Fuel Pin Mechanical Design

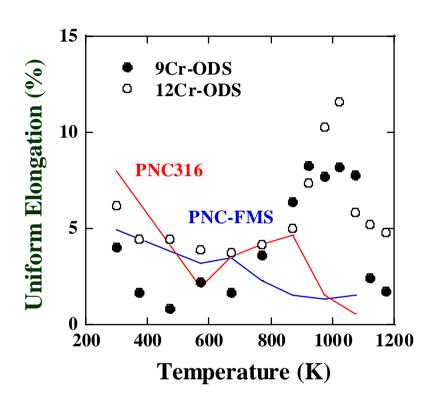
Material strength	1. Short-term tensile strength (σ_y , σ_u)	
	2. Creep rupture strength (σ_R)	
	3. Fatigue strength	
	4. Transient burst rupture strength	
Weld material strength	1. Short-term strength (σ_y , σ_u)	
	2. Creep rupture strength	
	3. High-cycle fatigue property	
Stress-strain correlation	1. Short-term stress-strain	
	2. Thermal creep strain	
	3. Irradiation creep strain	
Environmental effects	Environmental effect adjusting factor	
	2. Sodium corrosion	
	3. Fuel-to-cladding chemical interaction: FCCI	
	4. Void swelling	
Physical properties	1. Young modulus, 2. Poisson ratio, 3. Thermal expansion, 4. Thermal conductivity, 5. Heat capacity, 6. Density, 7. Transformation point	
Specifications	 Manufacturing process, 2. Chemical composition, Final heat treatments 	

JNC TN9400 2005-015 (2005)



Tensile Property



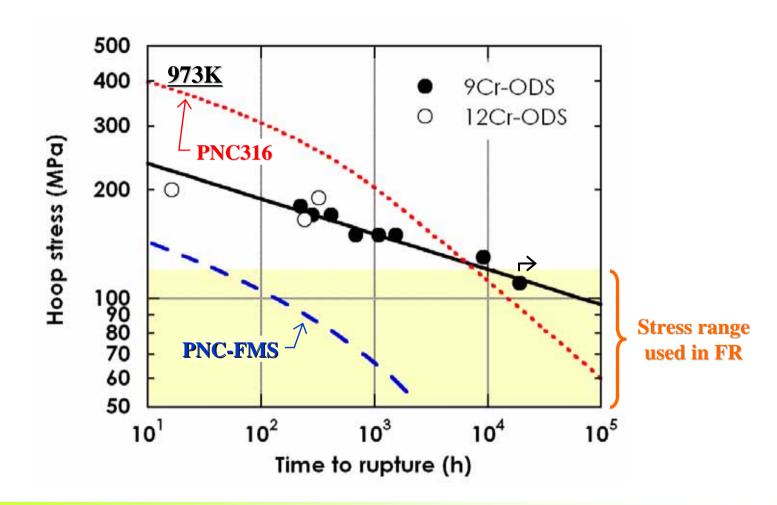


Ultimate tensile strength of 9Cr-ODS steel is much higher than precipitation hardened PNC-FMS.

Uniform elongation of both 9Cr- and 12Cr-ODS steels are comparable with PNC316 and PNC-FMS.



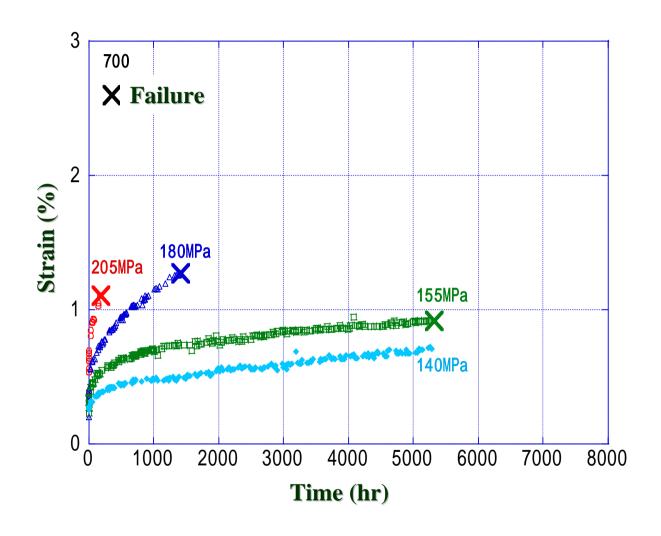
Creep Rupture Properties



Creep rupture strength of both 9Cr- and 12Cr-ODS steels under internal pressures meet the target: 120 MPa for 10,000 hr at 973 K.



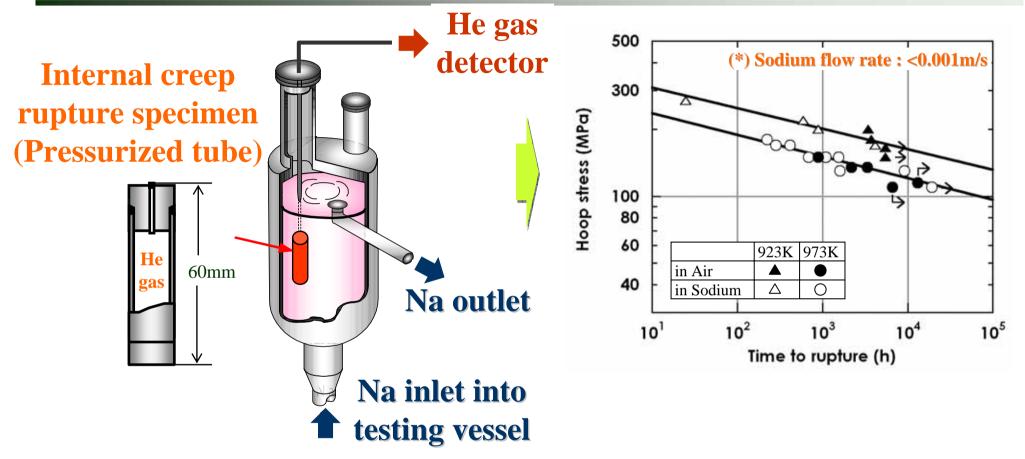
Creep Curves of 9Cr-ODS Steel



Creep curves under axial stress show little ternary stage in both 9Cr- and 12Cr-ODS steels



Sodium Immersion Effect on Creep Rupture Property



- -The 9Cr-ODS steel has superior sodium compatibility up to 973 K in stagnant sodium condition.
- -Decarburization and Ni intrusion will be more influential For the 9Cr-ODS steel in flowing sodium and in-pile conditions.



Chapter 6. Irradiation Tests

Step1. Material Specimen Irradiations

- Screening of candidate materials
- Fundamental investigation of irradiation behavior
- Material Strength Standards

Step2. Fuel Pin Irradiations

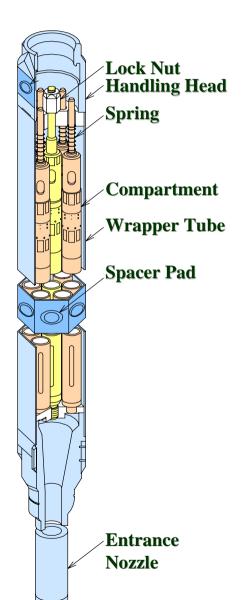
- Step 1 +
- Fuel to cladding mechanical/chemical interaction

Step3. Leading/Dedicated Subassembly Irradiations

- Demonstration

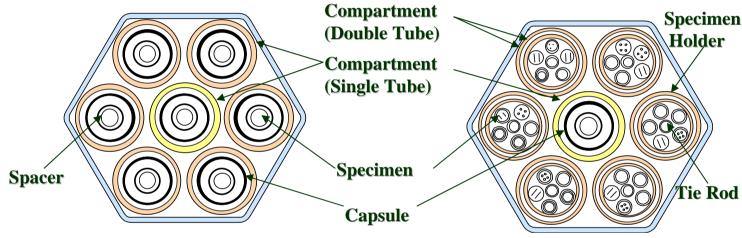
Material Irradiation Rig: Off-line Type - CMIR & SMIR -





Cross section of Structure Materials Irradiation Rig (SMIR)

Cross section of Core Materials Irradiation Rig (CMIR) Compartment





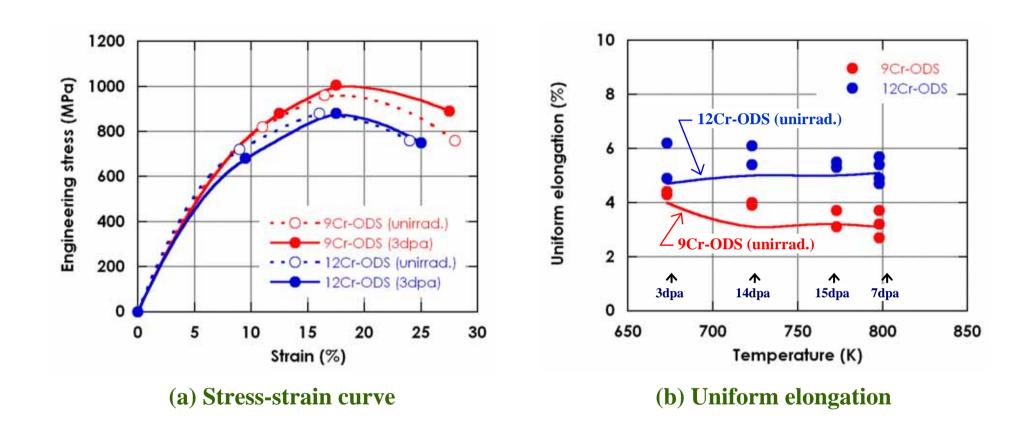
Example of Specimen and Capsule



Example of Miniature Specimen and Holder



Irradiation Effect on Tensile Property



Strength and ductility levels of irradiated ODS steel cladding tubes are adequately maintained.



Fuel Pin Irradiations in EBR-II

JAEA(PNC)-DOE Collaboration

12 Fuel Pins

- 1DK: 13Cr-3W-0.5Ti-0.35Y2O3-0.07Ex.O
- 1DS: 0.1C-11Cr-3W-0.5Ti-0.5Y2O3-0.1Ex.O
- Higher Smear Density MOX Fuel Pellets
- PRW

Irradiation: Nov.26, 1992- Sep.15, 1994

- Two subassembly: SPA-1B, C2
- LHR: ~48 kW/m, Tcm:~640
- Burnup:~6.3 at%, Fluence:~24 dpa

Post Irradiation Examinations

- Profilometry, Ceramography

Fuel Pin Irradiations in BOR-60 (1)



- JAEA-RIAR Collaborative Work -

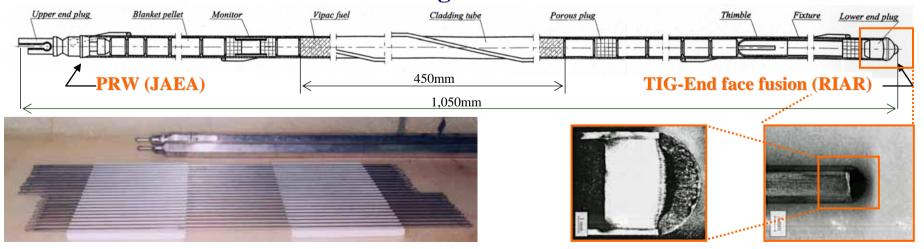
Objectives

- To investigate irradiation performance of ODS clad fuel pins
- To demonstrate and confirm the integrity up to 150 GWd/t

Specification of ODS Clad Fuel Pins

Fuel pin		Cladding	Engl	Dr./(Dr. LI)	Crease density
Outer diameter	Length	material	Fuel	Pu/(Pu+U)	Smear density
6.9 mm	1,050 mm	9Cr-ODS 12Cr-ODS	Vibro-packed MOX fuel	15 wt%	9.0 g/cm ³

Fuel Pin Structure and Manufacturing



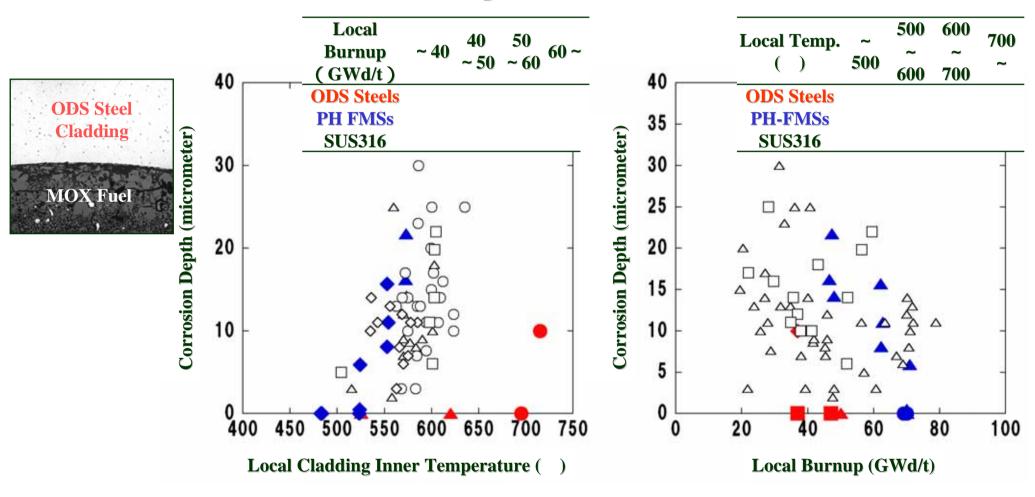
RIAR: Research Institute of Atomic Reactors

Fuel Pin Irradiations in BOR-60 (2)



- JAEA-RIAR Collaborative Work -

Fuel-to-Cladding Chemical Interaction

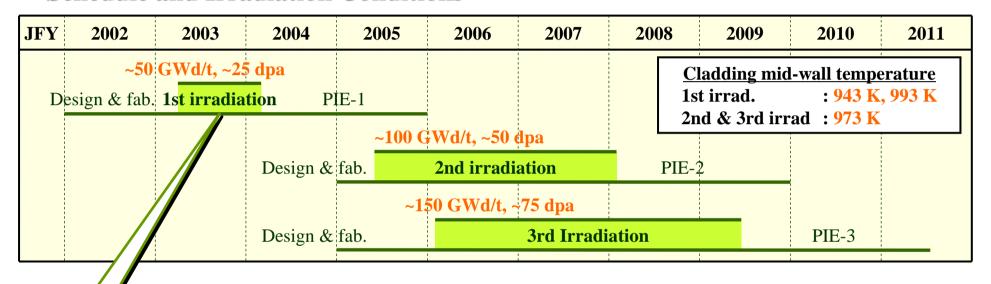


In case of BOR-60 irradiations, vibro-packed fuel contains uranium getter to control stoichiometry. This low stoichiometric condition (oxygen-to-metal ratio: ~1.93) will strongly influence (suppress) internal corrosion behavior

Fuel Pin Irradiations in BOR-60 (3) - JAEA-RIAR Collaborative Work -



Schedule and Irradiation Conditions



Results

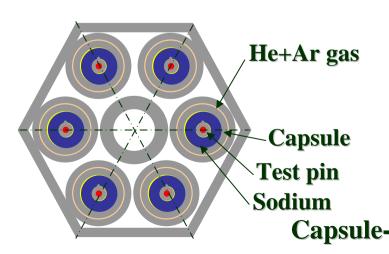
- Irradiation results of CDF up to 0.3 were achieved without fuel pin failure.
- Maximum corrosion depth observed was 10 micrometer which is comparable with PNC-FMS.

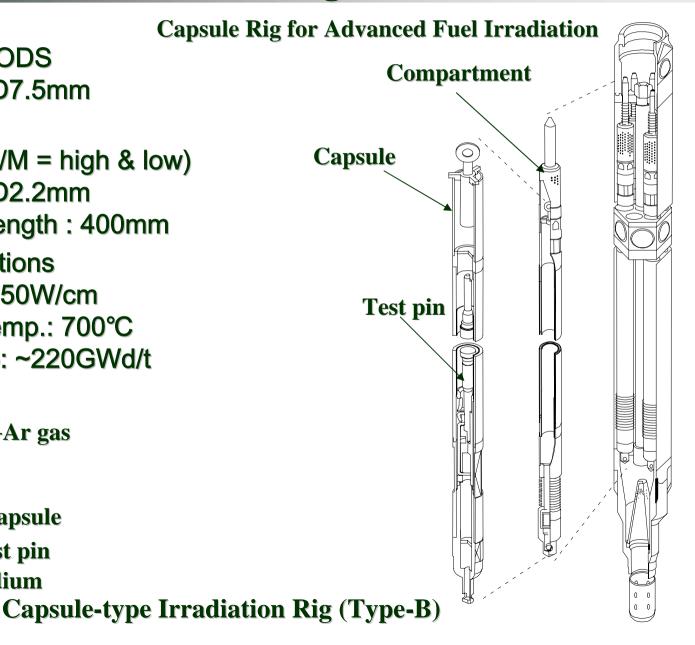
Fuel Pin Irradiations in JOYO



- Irradiation Rig -

- 1. Cladding tubes
 - Material: 9Cr-ODS
 - OD8.5mm x ID7.5mm
- 2. Fuel pellets
 - (U, Pu) O_{2-x} (O/M = high & low)
 - OD7.3mm x ID2.2mm
 - Fuel column length: 400mm
- 3. Irradiation conditions
 - Target LHR: 450W/cm
 - Target Clad temp.: 700°C
 - Target burnup: ~220GWd/t





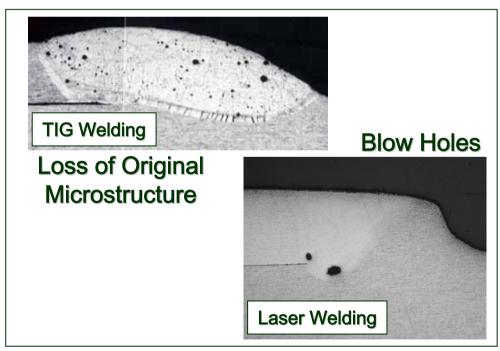


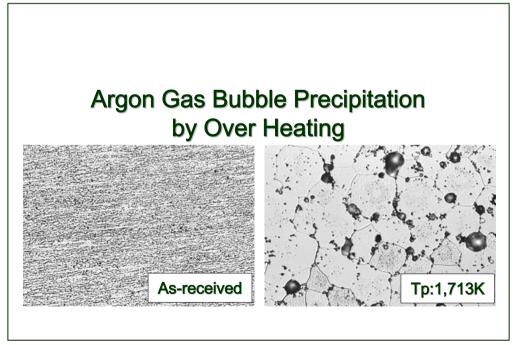
PRW Technology (1)

When ODS steels are heated far beyond 1,473 K (1,200), nano-meter size dispersoids will coarsen and lose dispersion strengthening effects and argon gas bubble will precipitate at grain boundary.

Pressurized Resistance Welding (PRW) Process*

*: Electrical resistance heating of the components while maintaining a continuous force sufficient to forge weld without melting.



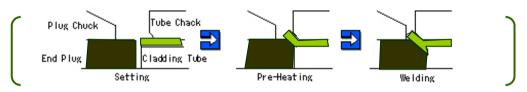


Photographs are taken by Fuel Technology Research and Development Section, Fuel Technology Department, Plutonium Fuel Development Center, Tokai Research and Development Center



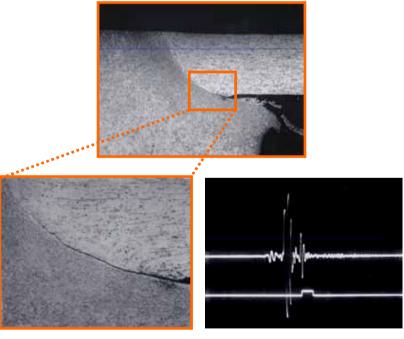
PRW Technology (2)

Welding Process









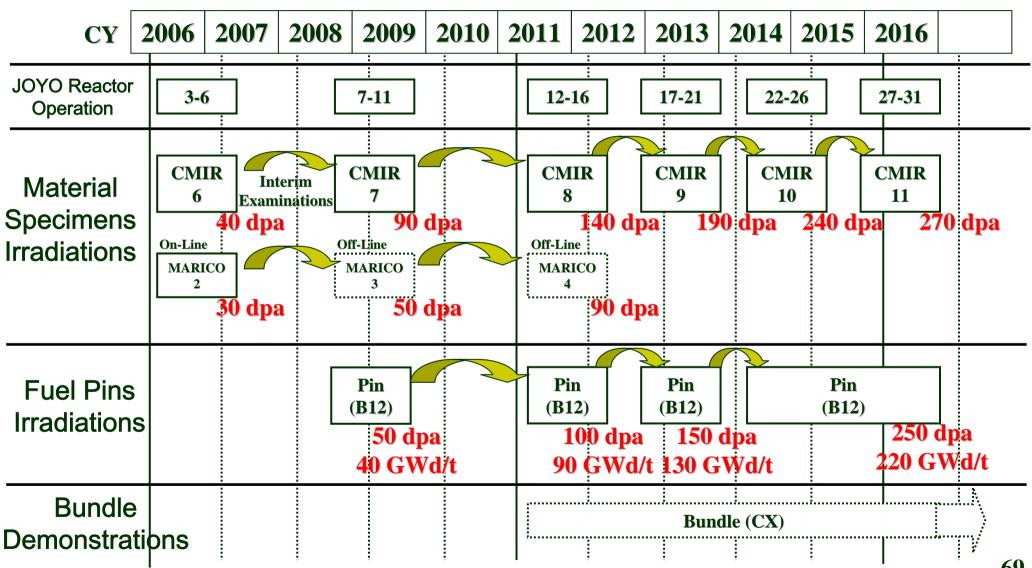
Ultrasonic Inspection

Available for fuel pin end-plugs and pressurized tube specimens



Schedule of JOYO Irradiation Tests

<Scheduled at beginning of FaCT project>





Chapter 7. Future Mass Production

Optimistic Estimation of Market Scale for ODS Steel Cladding Tubes

	CFBR	
Core Fuel Pin Geometry	OD8.5mm × L3,000mm	
Core Fuel Pins/Subassembly	255	
Fuel Subassembly	~800	
Operation Cycle Length	~26 month	
Number of the Tubes	~24,000 /year*	
Weight of the Tubes	~7,000 kg/year**	

^{*:} does not include loss in manufacturing and fuel pin fabrication processes.

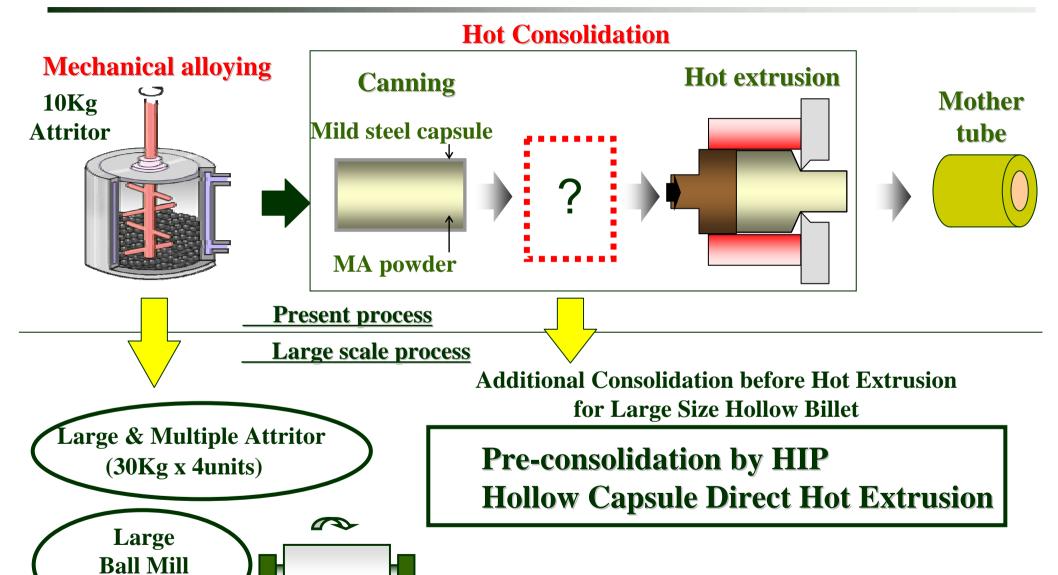
JOYO Irradiation Tests: 24 tubes/year (4.4kg)

Scale-up of factor 1,000 is necessary for commercialization!!70

^{**:} The weight of raw steel powders may be three times larger than that of products.



Key Process for Large Scale Production



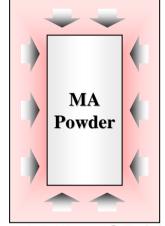
(>250 Kg)



Pre-Consolidation by HIP and Hot Extrusion

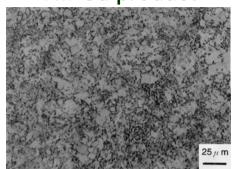








Optical micrograph of HIPed product





HIPed Billet: OD225mm×ID60mm ×L500mm: 150kg

Hot Extrusion Test of HIPed Billets@1,453K







Annular Billets: OD65mm×ID48mm ×L10,000mm:120kg 72



Direct Hot Extrusion by Hollow Capsule

Powder Filling



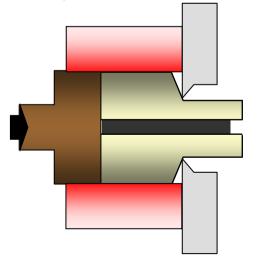
Assembling



OD147mm×ID32mm ×L590mm:30kg

Hot-Extrusion 2,000ton Press

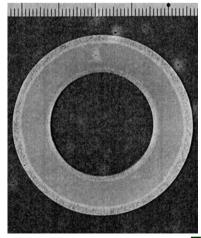
@1,423K, Extrusion Ratio 13





Annular Billets:OD32mm×ID21mm ×L2,000mm:7.3kg

High Chromium Steel Liner Cross Section





Conclusions: 1

Candidate

Martensitic 9Cr-ODS steel: Primary

Fully ferritic 12Cr-ODS steel: Secondary

Alloy Design: Strength, Ductility, Microstructure Control

- Phase and Grain Morphology Control: C,Cr, (Ex.O)
- Solution & Dispersion Hardening: W, Ti, Y₂O₃, Ex.O

Microstructure Control:

- Nano-size oxide particles precipitate during PM process
- Grain morphology must be controlled during tubing process
- 9Cr-ODS: Alpha to Gamma Transformation Cooling Rate Control
- 12Cr-ODS: Recrystallization Two Step Heat Treatment



Conclusions: 2

Mechanical Properties

- Tensile and Creep Rupture Strength met the Initial Target
- Environmental Effects

Manufacturing Process for Future Mass Production

- Screening Feasible Technologies

Larger Size Hot-Extruded Billets
Pre-consolidation by HIP
Direct Hollow Capsule Extrusion

Irradiation Tests

- Material Specimen Irradiations in JOYO in progress
- Fuel Pin Irradiation Tests in BOR-60 since 2003
- PRW technology



Activity in FaCT Project by 2015

Manufacturing Technology Development for Mass Production

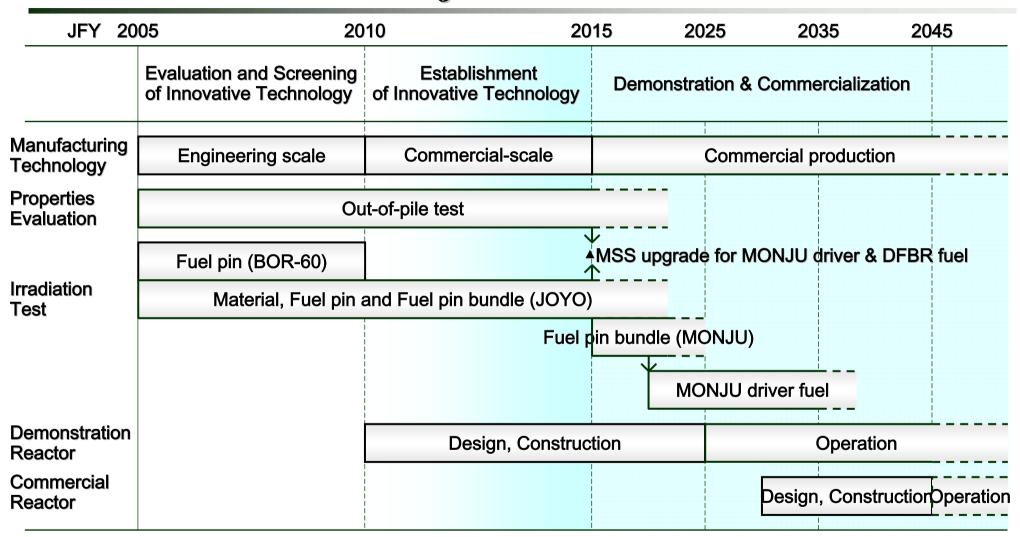
- Applicable for the annual production size:~10,000 tubes
- Larger Mother Tube Volume
- Longer Final Product
- Quality Assurance

Demonstration of High Burnup Capability

- Material Specimen Irradiations in JOYO up to 250 dpa
- Fuel Pin Irradiations in BOR-60 to 2009; 150GWd/t/75dpa
- Fuel Pin Irradiations in JOYO; ~180GWd/t/~210dpa
- Upgrading Material Strength Standard



Project Schedule



Evaluate and Select Innovative Technologies by 2010: ODS clad fuel pin data at 150GWd/t and evaluation of high burnup capability Establish Innovative Technologies by 2015: ODS clad fuel pin data at 250GWd/t and fuel pin bundle irradiation at medium burnup Demonstrate Innovative Technologies by 2025: Fuel pin bundle/subassembly demonstration in MONJU



Representative Publications

- [1] S. Ukai, S. Mizuta, M. Fujiwara, T. Okuda, T. Kobayashi, J. Nucl. Sci. Technol., Vol.39, No.7, pp.778-788 (2002).
- [2] J. Bottcher, S. Ukai, M. Inoue, Nucl. Technol., Vol. 138, pp. 238-245 (2002).
- [3] S. Ukai, H. Hatakeyama, S. Mizuta, M. Fujiwara, T. Okuda, J. Nucl. Mater., Vol. 307-311, pp. 758-762 (2002).
- [4] S. Ukai, T. Kaito, S. Ohtsuka, T. Narita, M. Fujiwara, T. Kobayashi, ISIJ International, Vol.43, No.12, pp.2038-2045 (2003).
- [5] T. Yoshitake, Y. Abe, N. Akasaka, S. Ohtsuka, S. Ukai, A. Kimura, J. Nucl. Mater., Vol.329-333, pp.342-346 (2004).
- [6] S. Ohtsuka, S. Ukai, M. Fujiwara, T. Kaito, T. Narita, J. Nucl. Mater., Vol.329-333, pp.372-376 (2004).
- [7] E. Yoshida, S. Kato, J. Nucl. Mater., Vol.329-333, pp.1393-1397 (2004).
- [8] M. Seki, K. Hirako, S. Kono, Y. Kihara, T. Kaito, S. Ukai, J. Nucl. Mater., Vol.329-333, pp.1534-1538 (2004).
- [9] T. Narita, S. Ukai, T. Kaito, S. Ohtsuka, T. Kobayashi, J. Nucl. Sci. Technol., Vol.41, No.10, pp.1008-1012 (2004).
- [10] S. Yamashita, N. Akasaka, S. Ohnuki, J. Nucl. Mater., Vol.329-333, pp.377-381 (2004).
- [11] S. Ukai, T. Kaito, M. Seki, A. A. Mayorshin, O. V. Shishalov, J. Nucl. Sci. Technol., Vol.42, No.1, pp.109-122 (2005).
- [12] T. Narita, S. Ukai, T. Kaito, S. Ohtsuka, M. Fujiwara, J. Nucl. Sci. Technol., Vol.42, No.9, pp.825-832 (2005).
- [13] T. Kaito, S. Ukai, S. Ohtsuka, T. Narita, GLOBAL2005, Paper No.169, October 9-13, 2005, Tsukuba, Japan (2005).
- [14] S. Ohtsuka, S. Ukai, H. Sakasegawa, M. Fujiwara, T. Kaito, T. Narita, Mater. Trans., Vol.46, No.3, pp.487-492 (2005)
- [15] S. Ohtsuka, S. Ukai, M. Fujiwara, T. Kaito, T. Narita, J. Phys. Chem. Solids., Vol.66, pp.571-575 (2005).
- [16] S. Ukai, T. Kaito, S. Ohtsuka, T. Narita, H. Sakasegawa, Transactions of the 2006 ANS Annual Meeting, June 4-8, 2006, Reno, USA (2006).
- [17] S. Ohtsuka, S. Ukai, H. Sakasegawa, M. Fujiwara, T. Kaito, T. Narita, J. Nucl. Mater., Vol.367-370, pp.160-165 (2007)
- [18] T. Kaito, S. Ohtsuka, M. Inoue, GLOBAL2007, Paper No.005_175716, September 9-13, 2007, Idaho, USA (2007).