

**[C9. C. Doussier-Brochard] Magnetic structure of an antimony manganese chlorosulfide MnSbS<sub>2</sub>Cl**

Relatively to chalcogenides or halogenides, mixed compounds of the halogeno-chalcogenide type have been poorly studied up to now. Among them, quaternary compounds combining a transition metal *TM* with another cation are of special interest for their physical properties, due to the dilution of *TM* in the crystal matrix and the competition of the two types of cations versus the two ligands.

The crystal structure of MnSbS<sub>2</sub>Cl can be described as edge-sharing MnS<sub>4</sub>Cl<sub>2</sub> octahedra along the *b*-axis, and corner-sharing along the *a*-axis, forming waved layers separated by Bi atoms. The magnetic susceptibility versus temperature, shows a large maximum, around 39 K, characteristic of a low-dimensional anti-ferromagnetic behaviour, following by an increase at 27K.

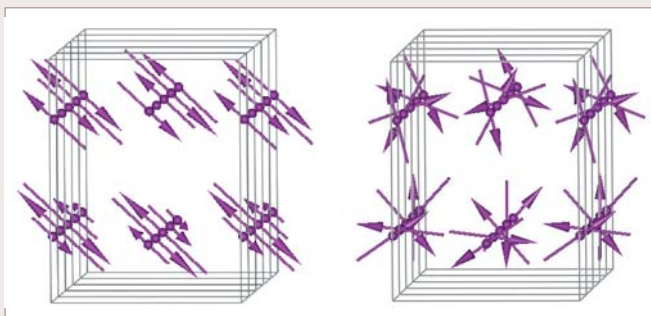
To elucidate this magnetic behaviour, powder neutron diffraction experiments were performed on the multi-detector G4.1 and allowed us to solve the magnetic structure and its thermal evolution [1]. Below T<sub>N</sub>=35 K, a complex 3D long-range antiferromagnetic ordering takes place, characterized by an incommensurate 1D propagation wave-vector along the *b*-axis, equal to  $k = [0, 0.3838, 0]$ . The Rietveld refinements give two possible modulation models, sinusoidal and helicoidal, with similar magnetic reliability factors and a moment of 4.5 μ<sub>B</sub> on Mn at 1.5 K.

MnSb<sub>2</sub>S<sub>4</sub> and MnPb<sub>4</sub>Sb<sub>6</sub>S<sub>14</sub>, which present chains of MnS<sub>4</sub> octahedra separated respectively by 6 and 16 Å, have been also studied to show the evolution of magnetic structure versus 2D or 1D organization [2].

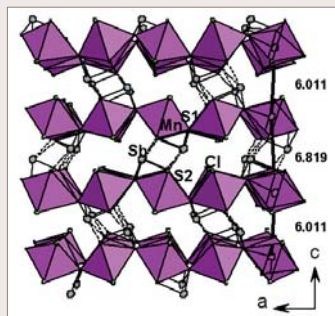
[1] C. Doussier, G. André, P. Léone, E. Janod, Y. Moëlo, Journal of Solid State Chemistry, 179 (2006) 486

[2] C. Doussier, thesis, Université de Nantes (2006)

[Collaboration : C. Doussier, P. Léone, E. Janod, Y. Moëlo Université de Nantes, G. André, LLB]



Magnetic structure with sinusoidal (left) or helicoidal (right) modulation



Crystal structure of MnSbS<sub>2</sub>Cl

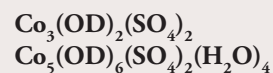
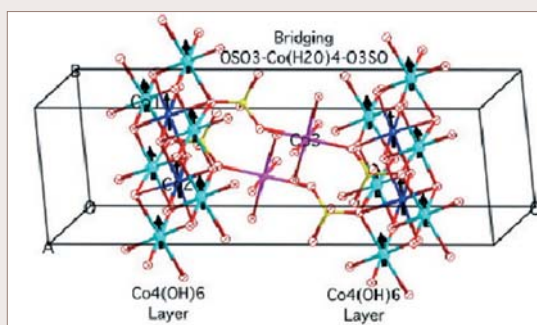
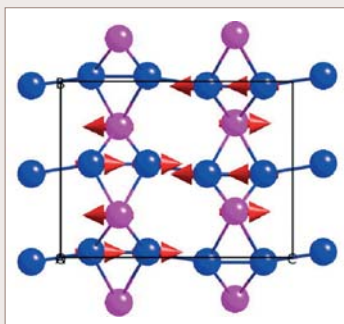
**[C10. S. Vilminot] Magnetic structures of the synthetic magnetic minerals based on hydroxysulfates of divalent metals, Co<sub>3</sub>(OD)<sub>2</sub>(SO<sub>4</sub>)<sub>2</sub>, Co<sub>5</sub>(OD)<sub>6</sub>(SO<sub>4</sub>)<sub>2</sub>(H<sub>2</sub>O)<sub>4</sub>, Cu<sub>4</sub>(OD)<sub>6</sub>SO<sub>4</sub>**

High-resolution (3T2) and high-flux (G4.1) neutron powder diffraction (NPD) data have been used to determine the nuclear and magnetic structures of a series of M<sup>II</sup>-hydroxysulfates, M = Mn, Co, Ni and Cu, as part of an ongoing project to understand short-range (SRO) and long-range (LRO) magnetic ordering and geometrical frustration caused by μ<sub>3</sub>-OH bridges. For the M<sub>3</sub>(OD)<sub>2</sub>(SO<sub>4</sub>)<sub>2</sub> series which are canted antiferromagnets [T<sub>N</sub> = 42 (Co), 26 (Mn) and 29 K (Ni)], the Ni exhibits collinear moments along *b* while for Co and Mn the moments lie in the *ac*-plane with a rare co-existence of SRO and LRO.[1]

Co<sub>5</sub>(OD)<sub>6</sub>(SO<sub>4</sub>)<sub>2</sub>(H<sub>2</sub>O)<sub>4</sub> consists of triangular Co<sup>II</sup>-OH layers pillared by ...O<sub>3</sub>SO-Co<sup>II</sup>(H<sub>2</sub>O)<sub>4</sub>-OSO<sub>3</sub>...and it behaves as a ferromagnet below 14 K.[2] Extensive magnetization by varying temperature, field and pressure and heat capacity measurements and NPD reveal an easy-plane XY-magnet where the moment of the pillaring Co remains random. Due to slight anisotropy in the layer the moments are oriented along *b*. These results demonstrate, for the first time, the existence of LRO in a single layer.

Cu<sub>4</sub>(OD)<sub>6</sub>SO<sub>4</sub> has a complex corrugated layered structure and it behaves as a canted AF. NPD was modelled with the moments oriented collinearly perpendicular to the corrugated planes with alternation along ±*a* for neighbouring chains within double chains building up the planes [3].

[Collaboration : S. Vilminot, IPCMS, Strasbourg; G. André, F. Bourée-Vigneron, LLB; M. Kurmoo, ULP, Strasbourg]



[1] M. Ben Salah *et al.*, *Chem. Mater.* 17 (2005) 2612-2621.

[2] M. Ben Salah *et al.*, *J. Am. Chem. Soc.* 128 (2006) 7972-7981.

[3] S. Vilminot *et al.*, *Dalton Trans.*, (2006) 1455-1462.